Cracking the Periodic Table Code

Why aren't the elements listed in alphabetical order?

Whyt

As charts go, the periodic table is a bit odd. It's not square. Large portions of the table appear to be missing at the top. It's not organized alphabetically so elements can be found easily. But so a clemist, the periodic table is a very powerful tool. The periodic table is organized by praperties, both chemical and physical. Those properties relate to the electronic structure of the atoms of each element. In today's activity, you are going to study how the ground state electron configurations and structure of atoms are selected to the shape and organization of the periodic table.

Model 1 - Blank Periodic Table

a Milack

1.	Te^{i}	Dr.									p-block			
2	25	20								11	55.79			
8	30	jk^{α}				a								
40	467	46		34"	30	34"	34"	38	34"					
8	50	50		4.0	440	440	44"	45	44"		5p-1	56"	590	
6	60	647									19-1	69-1	690	
7	297	29.5				333								

f-block.

* 1	401	40	40'00	:077	400	q^{m}	
-	35"	50	57"	59**	5900	S^{res}	

- 1. Obtain a card with electron configurations for your group as assigned by your reacher,
 - a. Bacond your coars number / and corn name Affail would
 - it If or each element on your card, write the last orbital notation appearing at the end of the configuration (the underlined portion).

Annexes will say for each seam. The assumes for Turm 1 are down below.

H H No 34 No 54

 What is similar about the last orbital notation appearing at the end of the configuration for each element in year ser?

Assume will may for each team. For Team 1, the configuration allows end in the

Answers To Cracking The Periodic Table Code

M Mosston

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