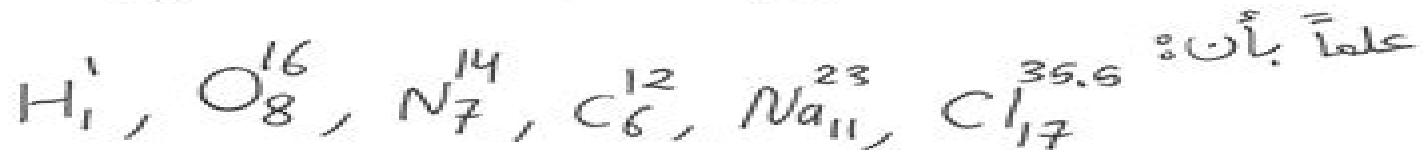


★ Molecular mass = الوزن الجزيئي "M.wt" 1

الوزن الجزيئي لأي مركب أو جزيء يساوي مجموع أوزان الذرات المكونة له.

e.g:



الحل

- $H_2 = H + H = 1 + 1 = 2 \text{ g/mol}$
- $H_2O = 2H + O = (2 \times 1) + 16 = 2 + 16 = 18 \text{ g/mol}$
- $NH_3 = N + 3H = 14 + (3 \times 1) = 14 + 3 = 17 \text{ g/mol}$
- $CH_4 = C + 4H = 12 + (4 \times 1) = 12 + 4 = 16 \text{ g/mol}$
- $CO_2 = C + 2O = 12 + [2 \times 16] = 12 + 32 = 44 \text{ g/mol}$
- $NaCl = Na + Cl = 23 + 35.5 = 58.5 \text{ g/mol}$

د. محمد عبد الله

Answers To Chemical Arithmetic

RM Cervero



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web answer 4 correct in the reactants and products must be the same part b what is the value of in the following nuclear reaction ${}^{65}_{29}\text{Cu} \rightarrow {}^{65}_{30}\text{Zn} + e^- + \bar{\nu}_e$ express your answer as an integer hint 1 how to approach the problem use rule 2 the total of the subscripts atomic numbers mathematically this can be written as answer

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