#### Post-Lab Questions:

 Which of the following compounds would form precipitates in an aqueous solution? NaCN, Na<sub>2</sub>CO<sub>3</sub>, FeSO<sub>4</sub>, ZnS, Ca(OH)<sub>2</sub>, CuCO<sub>3</sub>

#### Answer:

- a. NaCN (Sodium Cyanide) Soluble in water. No precipitate formed.
- b. Na<sub>2</sub>CO<sub>2</sub> (Sodium Carbonate) Soluble in water. No precipitate formed.
- c. FeSO: (Iron (II) Sulfate) Soluble in water. No precipitate formed.
- d. ZnS (Zinc Sulfide) Insoluble in water. Precipitate formed.
- e. Ca(OH); (Calcium Hydroxide) Insoluble in water. Precipitate formed.
- f. CuCO<sub>2</sub> (Copper (II) Carbonate) Insoluble in water. Precipitate formed.
- .. Therefore, ZnS, Ca(OH) 2, and CuCO2 are the compounds that would form precipitates in an aqueous solution.
- 2. Define an aqueous solution.

Answer. An aqueous solution is a solution in which water is the solvent. This simply means a solution that something has been dissolved in water.

<u>Answer</u>: The reaction of sodium hydroxide and sulfuric acid is known as a neutralization reaction.

4. What would be the net ionic equation for the reaction of sodium hydroxide and sulfuric acid?

Answer: The net ionic equation is  $2Na^*(aq) + 2OH^*(aq) + 2H^*(aq) + SO_4^2^*(aq) - 2Na^*(aq) + SO_4^2^*(aq) + 2H_2O(l)$ 

5. What are the spectator ions in the above reaction?

Answer. The spectator ions in the above reaction are Na\* and SO<sub>4</sub>\*. This is because they do not participate in the chemical reaction and remain in the solution in their initial forms.

What precipitate is formed from the reaction of barium chloride and silver nitrate?
 Answer: Prom the reaction of barium chloride and silver nitrate, a white precipitate is formed.

# **Answers To Chemthink Questions Precipitates Lab**

Siddappa N.Byrareddy

**Answers To Chemthink Questions Precipitates Lab:** 

The book delves into Answers To Chemthink Questions Precipitates Lab. Answers To Chemthink Questions Precipitates Lab is a vital topic that must be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Answers To Chemthink Questions Precipitates Lab, encompassing both the fundamentals and more intricate discussions.

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