

ChemActivity 1

The Nuclear Atom

(What Is an Atom?)

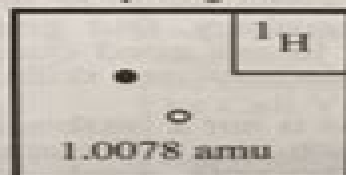
Model: Schematic Diagrams for Various Atoms and Ions.

- electron (-)
- proton (+)
- neutron (no charge)

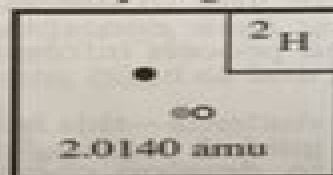
$$1 \text{ amu} = 1.6606 \times 10^{-24} \text{ g}$$

The **nucleus** of an atom contains the protons and the neutrons.

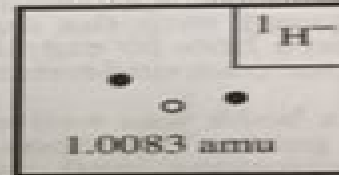
Hydrogen



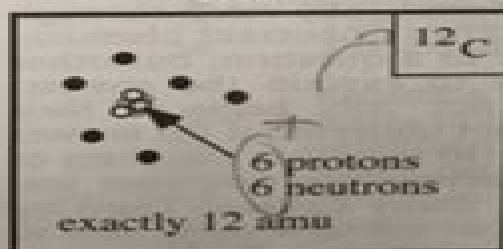
Hydrogen



Hydrogen ion



Carbon



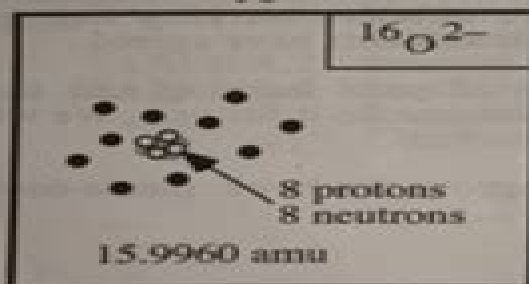
Carbon



Carbon ion



Oxygen ion



Sodium ion

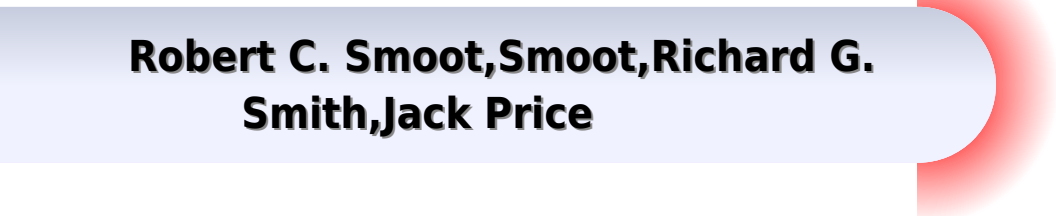


^1H and ^2H are **isotopes** of hydrogen.

^{12}C and ^{13}C are **isotopes** of carbon.

Answers To Nuclear Atom Chemactivity 1

**Robert C. Smoot, Smoot, Richard G.
Smith, Jack Price**



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Chemistry Richard S. Moog, John J. Farrell, 2014-01-13 Chemistry A Guided Approach 6th Edition follows the underlying principles developed by years of research on how readers learn and draws on testing by those using the POGIL methodology This text follows inquiry based learning and correspondingly emphasizes the underlying concepts and the reasoning behind the concepts This text offers an approach that follows modern cognitive learning principles by having readers learn how to create knowledge based on experimental data and how to test that knowledge **Introductory Chemistry** Michael P.

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Prentice Hall Chemistry, 2000 Merrill Chemistry Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998

Process Oriented Guided Inquiry Learning (POGIL) Richard Samuel Moog, 2008 POGIL is a student centered group learning pedagogy based on current learning theory This volume describes POGIL s theoretical basis its implementations in diverse environments and evaluation of student outcomes *Chemical Abstracts*, 1994-03-07 **Biological Abstracts**, 1962 Explaining the Atom Selig Hecht, 1947 This is a book for the complete layman It follows the steps that were taken historically in going from the earliest questions about the nature of common substances to the large scale liberation of atomic energy The number of these steps is small and the ideas involved are simple They can be described in all their essentials without assuming any knowledge of physics chemistry and mathematics Preface Atomic Theory and Structure of the Atom T. A. H. Peacocke, 2013-10-22 Atomic and Nuclear Chemistry Volume 1 Atomic Theory and Structure of the Atom presents the modern ideas of the atomic theory and atomic structure against the background of their historical development Topics covered include the classification of elements atoms and electrons the wave mechanical model of the atom and the determination of atomic weights This volume is comprised of six chapters and begins by discussing the origin of the atomic theory focusing on the role of John Dalton Avogadro s hypothesis and the introduction to the laws of chemical combination The chapters that follow look at the work of the early scientists that led to the development of the periodic table of elements

the use of the Avogadro number to determine the actual masses of atoms and molecules and the structure of the atom The essential results of the simple wave mechanical treatment are summarized in the next chapter This book concludes by considering developments in the determination of atomic weights Some brief notes on the character and personality of the great scientists who are mentioned throughout the text are included This book is intended for students and practitioners in the fields of chemistry and physics

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