

Date Monday 10/18/2021 Name Aya Turner
Section _____ Team _____
Instructor _____

Pre-Lab Study Questions

1. What is meant by the simplest formula of a compound?

The lowest whole number ratio of the atoms in the formula.

2. How does a hydrate differ from an anhydrate?

Hydrate is an ionic compound with water and anhydrate is an ionic compound without water.

3. What happens when a hydrate is heated?

When a hydrate is heated, it has the water removed and becomes an anhydrate.

4. A hydrate of CoCl_2 with a mass of 6.00 g is heated strongly. After cooling, the mass of the anhydrate is 3.27 g.

- a. How many grams of H_2O were lost from the hydrate?

$$\begin{array}{r} 6.00 \\ - 3.27 \\ \hline 2.73 \end{array} \quad 2.73 \text{ g } \text{H}_2\text{O}$$

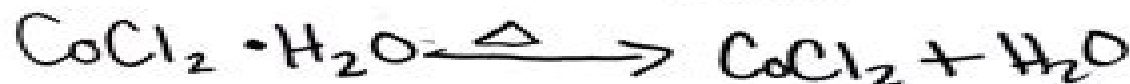
- b. What is the % water in the hydrate?

$$\frac{2.73}{6.00} \times 100 = 45.5\% \text{ of } \text{H}_2\text{O}$$

- c. What is the formula of the CoCl_2 hydrate?



- d. Write the equation for the dehydration of the CoCl_2 hydrate.



Answers To Chem Lab 11

S Ben Porath



Answers To Chem Lab 11:

Answers To Chem Lab 11 Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Answers To Chem Lab 11**," published by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect our existence. Throughout this critique, we will delve in to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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