
Chapter 2 - The Chemistry of Life exam study guide questions and answers

What three subatomic particles make up atoms? - answer The subatomic particles that make up atoms are protons, neutrons, and electrons.

Atom - answer The basic unit of matter.

Nucleus - answer The center of the atom that is formed by protons and neutrons bound together with strong forces.

Electron - answer A negatively charged particle (e^-) with only $1/1840$ mass of a proton that surrounds the nucleus.

How are all of the isotopes of an element similar? - answer Because they have the same number of electrons, all isotopes of an element have the same chemical properties.

Element - answer A pure substance that consists entirely of one type of atom.

Isotope - answer Atoms of the same element that differ in the number of neutrons they contain.

In what ways do compounds differ from their component elements? - answer The physical and chemical properties of a compound are usually very different from those of the elements from which it is formed.

Compound - answer A substance formed by the chemical combination of two or more elements in definite proportions.

What are the main types of chemical bonds? - answer The main types of chemical bonds are ionic bonds and covalent bonds.

Ionic Bond - answer Formed when one or more electrons are transferred from one atom to another.

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