

ChemThink: Behavior of Gases

Name _____ Per _____

1. What does the speed of atoms depend on? Describe the relationship between speed and each of these factors.

temperature: $\uparrow$ temp = $\uparrow$ speed

type of atom: smaller/lighter = faster

2. In a sample of one type of atom, do all the atoms move at the same speed? Why?

No, speed is related to each individual atoms energy

3. What is the equation for Pressure and what are the units for pressure?

$$\text{Pressure} = \frac{\text{Force}}{\text{Area}} \quad \text{psi}$$

4. When atoms are moving in a closed container what is creating the force on the sides of the container?

impacts (collisions) of atoms w/ sides of container

5. Explain how a change in temperature causes pressure to change and include an equation.

more collisions results in higher pressure

$\downarrow$
b/c molecules moving faster

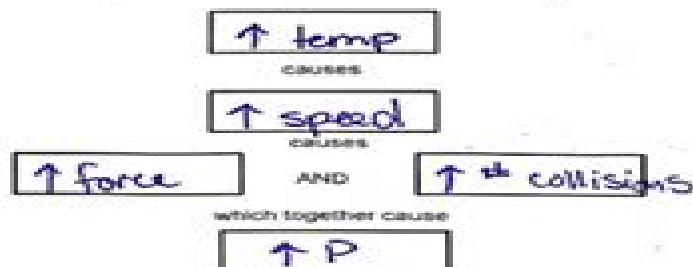
$$\frac{P}{T} = \text{constant}$$

6. Draw the graph you observed of pressure vs. temperature. Is this a direct relationship or an inverse relationship?



direct

7. Fill the cause/effect relationship in the diagram of boxes you observed.



- Explain how the number of atoms present causes pressure to change and include an equation?

more atoms = more collisions, which increases pressure

$$\frac{P}{n} = \text{constant}$$

Answers To Chemthink Gases

Jianjun Gao



Answers To Chemthink Gases:

Gases, 1984 *Gas Solubilities* William Gerrard, 2013-10-22 *Gas Solubilities Widespread Applications* discusses several topics concerning the various applications of gas solubilities The first chapter of the book reviews Henry's law while the second chapter covers the effect of temperature on gas solubility The third chapter discusses the various gases used by Horvut and the following chapters evaluate the data on sulfur dioxide chlorine data and solubility data for hydrogen sulfide Chapter 7 concerns itself with solubility of radon thoron and actinon Chapter 8 tackles the solubilities of diborane and the gaseous hydrides of groups IV V and VI of the periodic table Chapter 9 discusses the solubility of gases containing fluorine while Chapter 10 talks about Hildebrand's theory in the light of all gas solubility data Chapter 11 covers the hydrogen halide system while Chapter 12 deals with the solubility of gases in water and aqueous solutions of salts inorganic acids and bases and organic compounds Chapter 13 discusses gases in sea water while Chapter 14 covers aerosol propellants and Chapter 15 tackles the solubility of nitric oxide Chapter 16 discusses the biotechnological aspects and Chapter 17 talks about more on making holes Chapter 18 covers the evaluation of data on phosphine The book would be of great interest to researchers and professionals concerned with applications of the soluble nature of gases *The Specific Heats of Gases* James Riddick Partington, William George Shilling, 1924 *An Introduction to the Gas Phase* Claire Vallance, 2017-11-23 *An Introduction to the Gas Phase* is adapted from a set of lecture notes for a core first year lecture course in physical chemistry taught at the University of Oxford The book is intended to give a relatively concise introduction to the gas phase at a level suitable for any undergraduate scientist After defining the gas phase properties of gases such as temperature pressure and volume are discussed The relationships between these properties are explained at a molecular level and simple models are introduced that allow the various gas laws to be derived from first principles Finally the collisional behaviour of gases is used to explain a number of gas phase phenomena such as effusion diffusion and thermal conductivity **Gas Analysis** Louis Munroe Dennis, 1913 *Gases* Robin Nelson, 2005-01-01 A very simple introduction to what gases are giving examples of some different kinds **Gases**, 1970

Reviewing **Answers To Chemthink Gases**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**Answers To Chemthink Gases**," an enthralling opus penned by a very acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve into the book's central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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