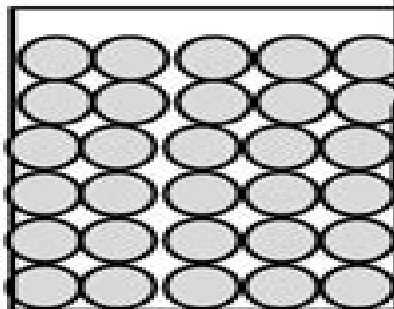


Name: _____

States of Matter

Directions: connect the phrase to it proper place.

Solid

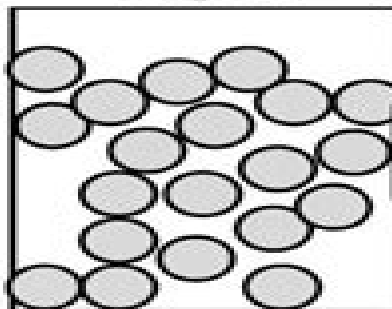


What is the shape of a solid?

What are some examples of solid?

What type of volume is solid?

Liquid

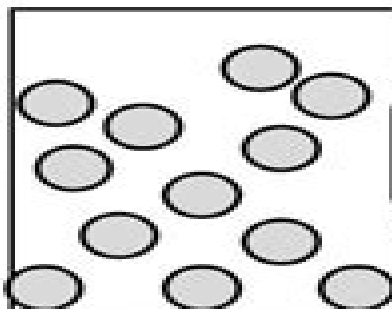


What is the shape of liquid?

What are some examples of a liquid?

What type of volume is a liquid?

Gas



What is the shape of a gas?

What are some examples of gas?

What type of volume is gas?

Phrase Bank:

Shape does not
change

Lemonade, soda,
paint, soap

Takes the shape of its
container

Ice cube, pencil, pa-
per, robot, dog

Takes the shape of its
container

Volume does not
change

Volume changes

Oxygen, helium, car-
bon dioxide

Volume does not
change

Answers To States Of Matter Review Packet

L Towne



Answers To States Of Matter Review Packet:

The Impact of State and National Standards on K-12 Science Teaching Dennis W. Sunal, Emmett L. Wright, 2006-06-01 This book addresses the expectations toward the science standards of various stakeholders including students parents teachers administrators higher education science and science education faculty members politicians governmental and professional agencies and the business community This book also investigates how the science standards have been translated into practice at the K 12 school district level addressing issues around professional development curriculum assessment evaluation and accountability The fundamental questions to be addressed are 1 What is the response in terms of trends and patterns of the educational system to the introduction of the national and state science standards since the late 1980 s and 2 What is the impact of the introduction of the science standards on teachers classrooms and students

The Nature of Matter Gr. 5-8 , Physical Science Grade 8 Bellaire, Tracy, 2013 Students learn about the development of western Canada from many perspectives Canadian government Aboriginals Metis and early immigrants They understand the contributions made by different individuals and groups and learn about the conflict and changes that occurred in the 19th century Includes 19 complete lesson plans with discussion questions for the topic reading passage and follow up worksheets and answer key , **Implementation of the Existing Particulate Matter and Ozone Air Quality Standards** United States. Congress. Senate. Committee on Environment and Public Works. Subcommittee on Clean Air, Climate Change, and Nuclear Safety, 2008 *Lewis, Newell, and Abbey Nominations* United States. Congress. Senate. Committee on Energy and Natural Resources, 2009 **Hearings held May 16** United States. Congress. House. Committee on the Judiciary, 1975 *FCC Record* United States. Federal Communications Commission, 1990 **Action of the governors under 39 U.S.C., section 3625, and supporting record in the matter of postal rate and fee increases, 1971** United States. Postal Rate Commission, 1972 *The Daily Washington Law Reporter* , 1887 Vols for 1902 include decisions of the District of Columbia Court of Appeals and various other courts of the District of Columbia

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limited the company held at callendar house on 7th november 2019 at 09 30am david white dw chaired the meeting throughout no members of the public attended

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 $2 \text{ CO}_2(\text{g}) + \text{H}_2\text{O}(\text{g}) + 2 \text{ C}_2\text{H}_2(\text{g}) \rightarrow 2 \text{ H}_2\text{O}(\text{g}) + 2 \text{ C}_2\text{H}_6(\text{g})$ $\Delta H = 94.5 \text{ kJ}$
 $\text{H}_2\text{O}(\text{g}) + \frac{1}{2} \text{O}_2(\text{g}) \rightarrow \text{H}_2\text{O}(\text{l})$ $\Delta H = 71.2 \text{ kJ}$
 $\text{C}_2\text{H}_6(\text{g}) + \frac{7}{2} \text{O}_2(\text{g}) \rightarrow 2 \text{ CO}_2(\text{g}) + 3 \text{ H}_2\text{O}(\text{g})$ $\Delta H = 283 \text{ kJ}$
 $\text{C}_2\text{H}_6(\text{g})$

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 $\text{C}_2\text{H}_4(\text{g}) + 3 \text{ O}_2(\text{g}) \rightarrow 2 \text{ CO}_2(\text{g}) + 2 \text{ H}_2\text{O}(\text{l})$ $\Delta H = 1411 \text{ kJ}$
 $\text{C}_2\text{H}_6(\text{g}) + 7 \text{ O}_2(\text{g}) \rightarrow 2 \text{ CO}_2(\text{g}) + 3 \text{ H}_2\text{O}(\text{l})$ $\Delta H = 1560 \text{ kJ}$
 $\text{H}_2(\text{g}) + \frac{1}{2} \text{O}_2(\text{g}) \rightarrow \text{H}_2\text{O}(\text{l})$ $\Delta H = 285 \text{ kJ}$
 mole

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process is the sum of the Δh values for each step to use hess s law two principles must be understood one if an equation is reversed the sign of the Δh value is also reversed two if an equation is multiplied by a

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web thus for reaction 9 2 in which 1 mole of H_2O and 1 mole of OH^- combine to form 2 moles of H_2O $\Delta H = -55.8 \text{ kJ}$ as shown note that the enthalpy change $\Delta H_{\text{products}} - \Delta H_{\text{reactants}}$ is positive if heat is absorbed that is if $\Delta H_{\text{products}} > \Delta H_{\text{reactants}}$

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