

Chapter 18- Radioactivity and Nuclear Reactions

I. Radioactivity

- A. The nucleus of an atom is composed of protons and neutrons which comprise most of the atom's mass.
- B. The strong force causes protons and neutrons to be attracted to each other in the nucleus.
 - 1. The strong force is powerful only when neutrons and protons are closely packed together.
 - 2. The protons and neutrons in a large nucleus are held less tightly by the strong force than protons and neutrons in a small nucleus.
- C. Radioactivity—nuclear decay which happens when the strong force is not large enough to hold the nucleus together; the nucleus gives off matter and energy.
 - 1. Isotopes are atoms of the same element with varying numbers of neutrons.
 - 2. A nucleus with too many or too few neutrons compared to protons is radioactive.
 - 3. Atomic number—number of protons in an atom
 - 4. Mass number—number of protons and neutrons in a nucleus
- D. Uranium radioactivity was discovered in 1896 by Henri Becquerel; Marie and Pierre Curie discovered the radioactive elements polonium and radium in 1898.

II. Nuclear Decay

- A. Nuclear radiation—particles and energy are released from a decaying nucleus.
- B. Alpha particle—two protons and two neutrons with an electric charge of +2
 - 1. Alpha particles leave charged ions in their path when they travel through matter.
 - 2. Alpha particles are the least penetrating form of nuclear radiation.
 - 3. Alpha particles can cause serious biological damage.
- C. Beta particle—neutron decays into a proton and releases an electron at high speed; more penetrating than alpha particles
- D. Gamma ray—penetrating electromagnetic waves that carry energy, but have no mass or charge
- E. Transmutation—process of one element's changing to another through nuclear decay
- F. The half-life of a radioactive isotope is the length of time it takes half the nucleus to decay; half-lives vary from fractions of a second to billions of years.
- G. Carbon dating can be used to date once-living materials while uranium dating can be used to date rocks.

III. Detecting Radioactivity

- A. Radiation detectors are instruments used to identify ions formed when radiation passes through matter.
 - 1. A cloud chamber detects alpha or beta particles by means of a trail of condensed vapor.
 - 2. A bubble chamber detects radioactive particles by means of a bubble trail in a superheated liquid.
 - 3. Electroscopes can measure charged particles in the air.
- B. A Geiger counter measures radioactivity by producing an electric current when radiation is present.
- C. Background radiation comes from radon gas produced in the Earth's crust, from cosmic rays, and from radioactive isotopes in the body.

Answers To Radioactivity Nuclear Reactions

Clare Smith



Answers To Radioactivity Nuclear Reactions:

Hearings United States. Congress Senate, 1959 *Cutnell & Johnson Physics* John D. Cutnell, David Young, Kenneth W. Johnson, Shane Stadler, 2022 The newly revised Twelfth Edition of Cutnell's Physics delivers an effective and accessible introduction to college and university physics It contains easy to follow explanations of critical math and problem solving concepts From kinematics to work and energy temperature heat electricity magnetism and optics as well as foundational concepts in more advanced subjects like special relativity Physics is the ideal introductory text for students from any background The greatest strength of the text is the synergistic relationship it develops between problem solving and conceptual understanding The book lays emphasis on building relevance of physics in day to day living and highlights the physics principles that come into play A wide range of applications that are biomedical in nature and others that deal with modern technology Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone, Theodore O.

Dolter, 2012-01-03 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections assessment exercises at the end each section and relevant chapter problems at the end of each chapter A new Math Check allows quick access to the needed basic skill The first chapter now includes brief introductions to several fundamental chemical concepts and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem Every concept in the text is clearly illustrated with one or more step by step examples Making it Real essays have been updated to present timely and engaging real world applications emphasizing the relevance of the material they are learning This edition continues the end of chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter **Nuclear Science Abstracts**, 1973

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Environmental Physics Clare Smith, 2001 First Published in 2002 Environmental Physics is a comprehensive introduction to the physical concepts underlying environmental science The importance and relevance of physics is emphasised by its application to real environmental problems with a wide range of case studies Applications included cover energy use and production global climate the physics of living things radioactivity environmental remote sensing noise pollution and the physics of the Earth The book makes the subject accessible to those with little physics background keeping mathematical treatment straightforward The text is lively and informative and is supplemented by numerous illustrations photos tables of useful data and a glossary of key terms

ESSENTIALS OF PHYSICS SANJAY MORESHWAR WAGH, DILIP ABASAHEB DESHPANDE, 2012-09-27 Physics is our attempt to conceptually grasp all the happenings around us Then realizing that concepts are the free creations of the human mind helps us develop proper understanding of a subject especially during formative stages This introductory book on Physics presents careful analysis of the developments of basic concepts for the beginners It is written in a way that stimulates students and creates a sustained interest in Physics so that studying the subject is enjoyable and satisfying The physical concepts are explained clearly enough for anyone to understand In this text the exercises are provided in three different categories namely as questions as problems and as multiple choice questions The first category of exercises contains thought provoking and descriptive questions The second category of exercises involves numerical computations The third category of exercises of multiple choice questions provides a reader with a flavour of the currently popular mode of examination Intended for the introductory level college physics courses the book will also be an invaluable resource for the students preparing for various competitive examinations

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