

Ions:	Solubility Rules:
Li^+ , Na^+ , K^+ , Rb^+ , Cs^+ , Fr^+ , NH_4^+	Always soluble
NO_3^- , $\text{C}_2\text{H}_3\text{O}_2^-$	Always soluble
Cl^- , Br^- , I^-	Insoluble with Ag^+ , Hg_2^{2+} , Pb^{2+} Soluble with everything else
SO_4^{2-}	Insoluble with Sr^{2+} , Ba^{2+} , Ag^+ , Hg_2^{2+} , Pb^{2+} Soluble with everything else
CO_3^{3-} , PO_4^{2-}	Soluble with Li^+ , Na^+ , K^+ , NH_4^+ Insoluble with everything else
OH^- , S^{2-}	Soluble with Li^+ , Na^+ , K^+ , NH_4^+ , Ca^{2+} , Sr^{2+} , Ba^{2+} Insoluble with everything else

Answers To Solubility Rules And Mystery Solutions

E Durkheim



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