

“PERIODIC TABLE PUZZLE” ANSWER KEY

1. **C** (Cannot be “I” because “I” is Hydrogen, which is a gas, not a metal.)
2. **F** (Group 2 is the Alkaline Earth Metal Family)
3. **A** (“Inactive” means *not reactive*, also known as “inert.” The Noble Gasses in Group 18 are inert because they do not easily react with other elements.)
4. **H** (“Active,” meaning “reactive,” Group 17, the Halogens, are the most reactive nonmetals. Letter “H” occupies the space of the element Fluorine, which is the most active nonmetal on the Periodic Table.)
5. **B** (“Semi-metal” is the same as a “metalloid,” referring to elements that have some of the physical and chemical properties typical of metals, but other properties that are not characteristic of metals.)
6. **D** (“E” is technically a Transition Metal, not an “Inner” Transition Metal)
7. **G** (Oxidation state, or oxidation number, refers to the charge an atom would have if it became an ion. Elements in Group 16, such as “G,” all have 6 valence electrons, meaning they would need 2 more electrons to complete their valence shell. Since electrons have negative charges, gaining 2 more would give the ion a charge of -2.)
8. **E** (Transition Metals like “E” do not follow the standard rules. For instance, many have more than one oxidation number. Also, answer can’t be “B” because B is a semimetal/metalloid, not a metal.)
9. **J** (“B” is in Group 13, meaning it has 3 valence electrons. It would be easier for it donate 3 negative electrons, leaving it with a charge of +3, than it would for it to gain 5, for a charge of -5.)
10. **I** (“I” is in the place of Hydrogen, which could gain 1 electron to fill the first electron shell, or lose 1 electron to become the common positive ion H^+ .)

Answer To Chemistry A Study Of Matter Periodic Puzzle

Rosina Ehmann



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Science Games and Puzzles, Grades 5 - 8 Schyrlet Cameron, Carolyn Craig, 2012-01-03 This book promotes science vocabulary building increases student readability levels and facilitates concept development through fun and challenging puzzles games and activities *Journal of Chemical Education*, 1925 Includes Report of New England Association of Chemistry Teachers and Proceedings of the Pacific Southwest Association of Chemistry Teachers Study Guide for Hein and Arena's Foundations of College Chemistry Peter Scott, Rachael Enriquez Porter, 1999-04-18 Deep Learning Stellan Ohlsson, 2011-01-31 Although the ability to retain process and project prior experience onto future situations is indispensable the human mind also possesses the ability to override experience and adapt to changing circumstances Cognitive scientist Stellan Ohlsson analyzes three types of deep non monotonic cognitive change creative insight adaptation of cognitive skills by learning from errors and conversion from one belief to another incompatible belief For each topic Ohlsson summarizes past research re formulates the relevant research questions and proposes information processing mechanisms that answer those questions The three theories are based on the principles of redistribution of activation specialization of practical knowledge and re subsumption of declarative information Ohlsson develops the implications of those mechanisms by scaling their effects with respect to time complexity and social interaction The book ends with a unified theory of non monotonic cognitive change that captures the abstract properties that the three types of change share *Student Study Guide to accompany Foundations of College Chemistry, 11th Edition* Morris Hein, Susan Arena, 2003-09-03 Work more effectively and gauge your progress along the way This Student Study Guide that accompanies Hein s Foundations of College Chemistry 11th Edition contains a review and self evaluation section challenge problems and answers and solutions for all exercises in the study guide Foundations of College Chemistry is the book that defined the prep intro chemistry market 35 years ago And it s been a market leader ever since Hein Arena is known for its accuracy clear no nonsense approach and direct writing style Strong problem solving and carefully constructed problem sets make this book a standout among its many imitators

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