

Stoichiometry Limiting Reactant

Answers

- (1) How many grams of calcium phosphate can be produced from the reaction between 2.50 L of 0.250 M calcium chloride with an excess of phosphoric acid?



$$2.50 \text{ L CaCl}_2 \times \frac{0.250 \text{ mol CaCl}_2}{1 \text{ L CaCl}_2} \times \frac{1 \text{ mol Ca}_3(\text{PO}_4)_2}{3 \text{ mol CaCl}_2} \times \frac{310 \text{ g Ca}_3(\text{PO}_4)_2}{1 \text{ mol Ca}_3(\text{PO}_4)_2} = 64.6 \text{ g Ca}_3(\text{PO}_4)_2$$

- (2) How many milliliters of 1.50 M nitric acid is required to react with 100.0 g of cuprous oxide?



$$100 \text{ g Cu}_2\text{O} \times \frac{1 \text{ mol Cu}_2\text{O}}{143 \text{ g Cu}_2\text{O}} \times \frac{14 \text{ mol HNO}_3}{3 \text{ mol Cu}_2\text{O}} \times \frac{1000 \text{ ml HNO}_3}{1.5 \text{ mol HNO}_3} = 2180 \text{ ml HNO}_3$$

- (3) Consider the following unbalanced reaction:



Balance the reaction. 3.25 g of NH_3 is allowed to react with 3.5 g of O_2 .

- (a) Which reactant is the limiting reagent?

$$3.25 \text{ g NH}_3 \times \frac{1 \text{ mol NH}_3}{17.03 \text{ g NH}_3} \times \frac{4 \text{ mol NO}}{4 \text{ mol NH}_3} \times \frac{30.01 \text{ g NO}}{1 \text{ mol NO}} = 5.73 \text{ g NO}$$

$$3.5 \text{ g O}_2 \times \frac{1 \text{ mol O}_2}{32 \text{ g O}_2} \times \frac{4 \text{ mol NO}}{5 \text{ mol O}_2} \times \frac{30.01 \text{ g NO}}{1 \text{ mol NO}} = 2.63 \text{ g NO}$$

O_2 is the limiting reagent.

- (b) How many grams of NO are formed?

2.63 g of NO are formed.

- (c) How much excess reagent remains after the reaction?

$$3.5 \text{ g O}_2 \times \frac{1 \text{ mol O}_2}{32 \text{ g O}_2} \times \frac{4 \text{ mol NH}_3}{5 \text{ mol O}_2} \times \frac{17.03 \text{ g NH}_3}{1 \text{ mol NH}_3} = 1.49 \text{ g NH}_3 \text{ used}$$

$$3.25 \text{ g NH}_3 - 1.49 \text{ g NH}_3 = 1.76 \text{ g NH}_3 \text{ remains.}$$

Answer Sheet To The Stoichiometry Limiting Reagent

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Chemistry John A. Olmsted, Gregory M. Williams, Robert Charles Burk, 2016-01-14 Olmsted Burk is an introductory general chemistry text designed specifically with Canadian professors and students in mind A reorganized Table of Contents and inclusion of SI units IUPAC standards and Canadian content designed to engage and motivate readers distinguish this text from many of the current text offerings It more accurately reflects the curriculum of most Canadian institutions Instructors will find the text sufficiently rigorous while it engages and retains student interest through its accessible language and clear problem solving program without an excess of material that makes most text appear daunting and redundant

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Cehmistry Textbook for College and University USA Ibrahim Sikder,2023-06-04 Cehmistry Textbook USA Applying Multiple-Reaction Stoichiometry to Chemical Reactor Modelling Guillermo Fernando Barreto,Carlos Daniel Luzi,2024-03-12 This book delves into the realm of Chemical Reaction Engineering CRE by showcasing the practical application of multiple reaction stoichiometry The authors critically assess various approaches commonly taught in undergraduate CRE courses to establish the relationships between changes in chemical species In doing so they propose an innovative conceptual alternative that is specifically tailored for undergraduate lectures The book carefully selects composition measures that effectively harness the power of stoichiometric relationships in elementary reacting systems and models which are typically covered in these courses Going beyond the basics it also offers a profound discussion on the value of chemical stoichiometry for tackling more intricate reaction systems and detailed models Moreover the book presents a simplified procedure that minimizes the reliance on complex linear algebra techniques making the book accessible to a wider range of readers

Multiple Solution Methods for Teaching Science in the Classroom Stephen DeMeo,2008 For the first time in science education the subject of multiple solution methods is explored in book form While a multiple method teaching approach is utilized extensively in math education there are very few journal articles and no texts written on this topic in science Teaching multiple methods to science students in order to solve quantitative word problems is important for two reasons First it challenges the practice by teachers that one specific method should be used when solving problems Secondly it calls into question the belief that multiple methods would confuse students and retard their learning Using a case study approach

and informed by research conducted by the author this book claims that providing students with a choice of methods as well as requiring additional methods as a way to validate results can be beneficial to student learning A close reading of the literature reveals that time spent on elucidating concepts rather than on algorithmic methodologies is a critical issue when trying to have students solve problems with understanding It is argued that conceptual understanding can be enhanced through the use of multiple methods in an environment where students can compare evaluate and verbally discuss competing methodologies through the facilitation of the instructor This book focuses on two very useful methods proportional reasoning PR and dimensional analysis DA These two methods are important because they can be used to solve a large number of problems in all of the four academic sciences biology chemistry physics and earth science This book concludes with a plan to integrate DA and PR into the academic science curriculum starting in late elementary school through to the introductory college level A challenge is presented to teachers as well as to textbook writers who rely on the single method paradigm to consider an alternative way to teach scientific problem solving

Master The NCERT for NEET Chemistry - Vol.1 2020
Arihant Experts, 2019-06-04 While beginning the preparation for Medical and Engineering Entrances aspirants need to go beyond traditional NCERT textbooks to gain a complete grip over it to answer all questions correctly during the exam The revised edition of MASTER THE NCERT based on NCERT Classes XI and XII once again brings a unique set of all kinds of Objective Type Questions for Physics Chemistry Biology and Mathematics This book Master the NCERT for NEET Chemistry Vol 1 based on NCERT Class XI is a one of its kind book providing 14 Chapters equipped with topic wise objective questions NCERT Exemplar Objective Questions and a special separate format questions for NEET and other medical entrances It also provides explanations for difficult questions and past exam questions for knowing the pattern Based on a unique approach to master NCERT it is a perfect study resource to build the foundation over NEET and other medical entrances

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Target IIT JEE (A complete solution in Chemistry) Class XI Umesh Jalan, 2021-04-16 The key notes and questions present in this book have been tested by millions of IIT JEE students over the years This book contains all the important and frequent ask concept which is drive from several notes an previous year paper of JEE AIPMT JIPMER AIIMS NEET and various state engineering and medical entrance examinations Even a below average student can crack JEE after doing this book

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