

Grade 11 Chemistry

INTERMOLECULAR FORCES & Physical Properties

EXAM QUESTIONS PART 3



NAME OF SUBSTANCE	FORMULA	BOILING POINT (°C)
Water	H ₂ O	100
Ethanol	CH ₃ CH ₂ OH	78
Bromine	Br ₂	58,8
Iodine	I ₂	184,3
Ammonia	NH ₃	-33,3
Phosphine	PH ₃	-87,7

Explain why ethanol is soluble in water. Refer to the relative strength of the intermolecular forces in ethanol and water.

Explain why the boiling point of iodine is higher than that of bromine. Refer to the intermolecular forces present in EACH substance in the explanation.

Answers For Grade 11 Intermolecular Formal Task

LL Leslie



Answeres For Grade 11 Intermolecular Formal Task:

Decoding **Answeres For Grade 11 Intermolecular Formal Task**: Revealing the Captivating Potential of Verbal Expression

In an era characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its ability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Answeres For Grade 11 Intermolecular Formal Task**," a mesmerizing literary creation penned by a celebrated wordsmith, readers attempt an enlightening odyssey, unraveling the intricate significance of language and its enduring affect our lives. In this appraisal, we shall explore the book is central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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