

Name the following acids

1. HCL hydrochloric acid
2. HClO₄ perchloric acid
3. HIO₃ iodic acid
4. HI hydroiodic acid
5. H₂SO₄ sulfuric acid
6. H₂S hydrosulfuric acid
7. HCN hydrocyanic acid
8. H₂CO₃ carbonic acid
9. HBrO₄ perbromic acid
10. HBrO₃ bromic acid
11. HC₂H₃O₂ acetic acid
12. H₃PO₃ phosphorous acid
13. H₃P hydrophosphoric acid
14. H₂CrO₄ chromic acid
15. H₂CrO₂ hypochromous acid
16. H₂Cr hydrochromic acid

Give formulas for the following acids

17. hydrofluoric acid HF
18. nitric acid HNO₃
19. nitrous acid HNO₂
20. hydroxic acid H₂O
21. periodic acid HIO₄
22. percarbonic acid H₂CO₄
23. hypobromous acid HBrO
24. bromous acid HBrO₂
25. permanganic acid HMnO₄
26. manganic acid HMnO₃
27. iodous acid HIO₂
28. sulfurous acid H₂SO₃
29. sulfuric acid H₂SO₄
30. perchloric acid HClO₄
31. chlorous acid HClO₂
32. acetic acid HC₂H₃O₂
33. phosphoric acid H₃PO₄
34. phosphorous acid H₃PO₃

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Chemistry Neil D. Jespersen, Alison Hyslop, 2021-11-02 Chemistry The Molecular Nature of Matter 8th Edition continues to focus on the intimate relationship that exists between structure at the atomic molecular level and the observable macroscopic properties of matter Key revisions in this edition focus on three areas The deliberate inclusion of more updated real world examples that relate common real world student experiences to the science of chemistry Simultaneously examples and questions have been updated to align them with career concepts relevant to the environmental engineering biological pharmaceutical and medical sciences Providing students with transferable skills with a focus on integrating metacognition and three dimensional learning into the text When students know what they know they are better able to learn and incorporate the material Providing a total solution through New WileyPLUS by fully integrating the enhanced eText with online assessment answer specific responses and additional practice resources The 8th edition continues to emphasize the importance of applying concepts to problem solving to achieve high level learning and increase retention of chemistry knowledge Problems are arranged in an intuitive confidence building order

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Table of Contents Answers For Naming Acids

1. Understanding the eBook Answers For Naming Acids
 - The Rise of Digital Reading Answers For Naming Acids
 - Advantages of eBooks Over Traditional Books
2. Identifying Answers For Naming Acids
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answers For Naming Acids
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answers For Naming Acids
 - Personalized Recommendations

- Answers For Naming Acids User Reviews and Ratings
- Answers For Naming Acids and Bestseller Lists
- 5. Accessing Answers For Naming Acids Free and Paid eBooks
 - Answers For Naming Acids Public Domain eBooks
 - Answers For Naming Acids eBook Subscription Services
 - Answers For Naming Acids Budget-Friendly Options
- 6. Navigating Answers For Naming Acids eBook Formats
 - ePub, PDF, MOBI, and More
 - Answers For Naming Acids Compatibility with Devices
 - Answers For Naming Acids Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answers For Naming Acids
 - Highlighting and Note-Taking Answers For Naming Acids
 - Interactive Elements Answers For Naming Acids
- 8. Staying Engaged with Answers For Naming Acids
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answers For Naming Acids
- 9. Balancing eBooks and Physical Books Answers For Naming Acids
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answers For Naming Acids
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Answers For Naming Acids
 - Setting Reading Goals Answers For Naming Acids
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Answers For Naming Acids
 - Fact-Checking eBook Content of Answers For Naming Acids

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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