2) Which of the following contain(s) polar co A) H <sub>2</sub> B) NaI C) Cl <sub>2</sub> 3) What is the formal charge and hybridization A) 0, sp <sup>2</sup> B) +1, sp <sup>2</sup> C) -1, sp  4) What of the following best describes the hyperides.	in the electron configuration of the OD) three E) four  valent bonds?  D) H2O E) both A and C  in for the highlighted Carbon?  () +1, sp E) 0, sp H3O  bridization of C in carbocation CH3	ator
A) none B) one C) two I  2) Which of the following contain(s) polar co  A) H <sub>2</sub> B) NaI C) Cl <sub>2</sub> 3) What is the formal charge and hybridization  A) 0, sp <sup>2</sup> B) +1, sp <sup>2</sup> C) -1, sp D  4) What of the following best describes the hy	valent bonds?  D) H2O E) both A and C  in for the highlighted Carbon?  () +1, sp E) 0, sp H3C  bridization of C in carbocation CH3	c—n
A) none B) one C) two I  2) Which of the following contain(s) polar co  A) H <sub>2</sub> B) NaI C) Cl <sub>2</sub> 3) What is the formal charge and hybridization  A) 0, sp <sup>2</sup> B) +1, sp <sup>2</sup> C) -1, sp D  4) What of the following best describes the hy	valent bonds?  D) H2O E) both A and C  in for the highlighted Carbon?  () +1, sp E) 0, sp H3C  bridization of C in carbocation CH3	c—n
A) H <sub>2</sub> B) NaI C) Cl <sub>2</sub> I  3) What is the formal charge and hybridizatio A) 0, sp <sup>2</sup> B) +1, sp <sup>2</sup> C) -1, sp E  4) What of the following best describes the hy	E) both A and C  for the highlighted Carbon?  () +1, sp E) 0, sp  bridization of C in carbocation CH;	c—h
3) What is the formal charge and hybridization A) 0, sp <sup>2</sup> B) +1, sp <sup>2</sup> C) -1, sp D 4) What of the following best describes the hy	n for the highlighted Carbon? H <sub>3</sub> (1) +1, sp E) 0, sp H <sub>3</sub> (1) bridization of C in carbocation CH;	c—N
A) 0, sp <sup>2</sup> B) +1, sp <sup>2</sup> C) -1, sp D  4) What of the following best describes the hy	()+1, sp E) 0, sp H <sub>3</sub> 0 bridization of C in <b>carbocation CH</b> ;	c-r
A) 0, sp <sup>2</sup> B) +1, sp <sup>2</sup> C) -1, sp D 4) What of the following best describes the hy	()+1, sp E) 0, sp H <sub>3</sub> 0 bridization of C in <b>carbocation CH</b> ;	c-r
	Taranta da a cara cara e menos menos asentes a caracter a recorde e conserva en conserva de conserva de la car	a± 0
	Taranta da a cara cara e menos menos asentes a caracter a recorde e conserva en conserva de conserva de la car	- Table 1
A) sp <sup>3</sup> B) sp <sup>2</sup> C) sp 1		2
	D) s	
5) what orbital is the nitrogen lone pair residir	so in the following molecule?	-
A) sp <sup>3</sup> B) sp <sup>2</sup> C) sp D) p	E)s	-
		N
<ol><li>6) Identify the most acidic carboxylic acid.</li></ol>		
	", " ", " L	
A B C	P OH FT OH	
Answer is E		
7) The conjugate acid of H <sub>2</sub> O is		
A) HO- B) H <sub>3</sub> O C) H <sub>3</sub> O+ I	O) H <sub>2</sub> O E) H <sub>2</sub> O <sup>+</sup>	
TO SERVICION - TO THE PARTY OF		
<ol><li>How many potential stereoisomers exist for</li></ol>	the following molecule.	1
A) 1 B) 2 C) 3 D) 4 E) 5	$\bigcup$	L.,
9) what is the name of the following compour	id?	
A) trans-4-ethylevelohexan-1-ol B) c/s-4	ethylhexan-1-ol	

10) which of the following step is part of the propagation in the methane chlorination?

A) CI-CI  $\longrightarrow$  2 CI · B)  $H_3C$  · + CI-CI  $\longrightarrow$   $H_3C-CI$  + CI · C)  $H_3C-H$  + · CI  $\longrightarrow$   $H_3C$  · + H-CI D) B and C

Answer is D

11) Based on the Hammond postulate, which of the following transition state best explains the following reaction step?

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Frank R. Spellman

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