

Calorimetry and Specific Heat Lab

Part 1; Introduction

Purpose- The purpose of this lab is to investigate how a specific heat substance can be determined using a coffee cup calorimeter.

Question- Using a calorimeter, how can we figure out a specific heat?

Hypothesis- If the metal has a low heat, the metal can be a good material for cooking because it will not need as much heat to raise the temperature of the metal in order to cook food.

Part 2; Materials

Thermometer	250 mL beaker	400 mL beaker	Ring stand	Two buret clamps
Water	analytical balance	Test-tube holder	Four test tubes	Hot plate
Lid for polystyrene cup	aluminum, copper, iron, and lead powder	50 mL graduated cylinder	Two polystyrene coffee cups	

Step 1; Gather materials

Step 2; Put together the coffee cup calorimeter.

- I Put one coffee cup in another and then put them in the 250ml beaker.
- I then Cut a hole in the center of the lid to hold our thermometer. Put the lid on the top cup.

Step 3: Measure Masses of the Metals

Answer Sheet For Calorimetry Specific Heat

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Standards**

A red circular graphic with a gradient, appearing as a partial circle or a thick arc, located to the right of the text box.

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web sep 26 2023 method make the egg mixture in a medium bowl whisk together the eggs milk and cinnamon stir in the orange zest and or soak the bread slices in egg mixture place each slice of bread into the milk egg mixture allowing the bread to soak in fry the french toast melt some butter in a large

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web aug 21 2023 ingredients $\frac{2}{3}$ cup milk 2 large eggs 1 teaspoon vanilla extract optional $\frac{1}{4}$ teaspoon ground cinnamon optional salt to taste 6 thick slices bread 1 tablespoon unsalted butter or more as needed

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web feb 23 2019 instructions whisk together egg mixture in a bowl vigorous whisk avoid cinnamon floating on top melt 15g 1 tbsp butter in a non stick skillet over medium heat dunk a piece of bread quickly into the egg mixture coating both sides place in pan repeat with more slices to fill cook for 2 $\frac{1}{2}$

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