

Unit: Chemical Bonding Answer Key "Covalent Bonding" - WS #3

Directions: Please answer each question using your notes from today's class.

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 What is the difference between a molecular formula, structural formula and an electron dot formula? Give an example of each.

A molecular formula gives the chemical make-up of one molecule. It does not show atom-to-atom connectivity. Example: C_2H_2

A structural formula uses lines instead of dots to show shared pairs of electrons.

Example: H-H

An electron dot formula uses dots instead of lines to show shared pairs of electrons. Example: H:H

2. What is the difference between single, double, and triple covalent bonds? How many shared electrons are in each?

The difference between the types of bonds is: a single bond is two shared electrons, a double is four shared electrons, and a triple is six shared electrons.

Draw the electron dot and structural formulas for the following molecules.

Substance	Rough copy (work!)	Electron dot	Structural
a) Cl ₂		Note: The electron dot structures have dots in between each atom instead of lines.	:çı—çı:
b) H ₂ S			:\$-н Н
c) Brl			ij <u></u> —₿ŗ:

Answers For Packet On Chemical Bonding

Jit Chakraborty & Asimesh Dutta Gupta and Ravikanth Kamlekar **Answers For Packet On Chemical Bonding:**

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web this trapping half court press can be effective against teams with mediocre inexperienced guards remember any trapping defense always carries some risk of giving up the easy weak side lay up initial set up diagram a two defenders are at the half court line x1 and x2 x3 and x4 are positioned near the three point arc

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web half court trap is a type of half court defense in which the defensive team attempts to disrupt set plays by the offensive team at the half court line in combination with a half court press the half court trap s sole purpose is to prevent the offensive team from employing any sort of set offense a typical half court trap involves one

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