

◀ Homework Chapter 6

Exercise 6.39: Problems - The Mole Concept

▼ Part A

How many carbon atoms are in a diamond (pure carbon) with a mass of 75 mg?

Express your answer using two significant figures.



$N = 2.5 \cdot 10^{21}$

C atoms



You have already submitted this answer. Enter a new answer.

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Answers Mastering Chemistry Homework

Orleans, Myron



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Research in Chemistry Education Liliana Mammino, Jan Apotheker, 2021-05-17 This volume emphasizes the role of chemical education for development and in particular for sustainable development in Africa by sharing experiences among specialists across the African continent and with specialists from other continents It considers all areas and levels of chemistry education gives specific attention to known major challenges and encourages explorations of novel approaches The chapters in this book describe new teaching approaches approach explorations and in class activities analyse educational challenges and possible ways of addressing them and explore cross discipline possibilities and their potential benefits for chemistry education This makes the volume an up to date compendium for chemistry educators and educational researchers worldwide

Chemistry Catherine E. Housecroft, Edwin C. Constable, 2010 This text integrates the three major branches of chemistry with the aim of enabling students to tackle more easily the problems within the subject and to apply chemistry to real life situations

EBOOK: GENERAL CHEMISTRY, THE ESSENTIAL CONCEPTS CHANG, 2013-01-07 EBOOK
GENERAL CHEMISTRY THE ESSENTIAL CONCEPTS *Ebook: Organic Chemistry* Janice Smith, 2014-10-16 Serious Science with an Approach Built for Today's Students Smith's Organic Chemistry continues to breathe new life into the organic chemistry world This new fourth edition retains its popular delivery of organic chemistry content in a student friendly format Janice Smith draws on her extensive teaching background to deliver organic chemistry in a way in which students learn with limited use of text paragraphs and through concisely written bulleted lists and highly detailed well labeled teaching illustrations Don't make your text decision without seeing Organic Chemistry 4th edition by Janice Gorzynski Smith

Digital Learning and Teaching in Chemistry Yehudit Dori, Courtney Ngai, Gabriela Szeinberg, 2023-07-12 Education is always evolving and most recently has shifted to increased online or remote learning Digital Learning and Teaching in Chemistry compiles the established and emerging trends in this field specifically within the context of learning and teaching in chemistry This book shares insights about five major themes best practices for teaching and learning digitally digital learning platforms virtual visualisation and laboratory to promote learning in science digital assessment and building communities of learners and educators The authors are chemistry instructors and researchers from nine countries contributing an international perspective on digital learning and teaching in chemistry While the chapters in this book span a wide variety of topics as a whole they focus on using technology and digital platforms as a method for supporting inclusive and meaningful learning The best practices and recommendations shared by the authors are highly relevant for modern chemistry education as teaching and learning through digital methods is likely to persist Furthermore teaching chemistry digitally has the potential to bring greater equity to the field of chemistry education in terms of who has access to quality learning and this book will contribute to that goal This book will be essential reading for those working in chemical education and teaching Yehudit Judy Dori is internationally recognised formerly Dean of the Faculty of Education of Science and

Technology at the Technion Israel Institute of Technology and won the 2020 NARST Distinguished Contributions to Science Education through Research Award DCRA for her exceptional research contributions Courtney Ngai and Gabriela Szeinberg are passionate researchers and practitioners in the education field Courtney Ngai is the Associate Director of the Office of Undergraduate Research and Artistry at Colorado State University Gabriela Szeinberg serves as Assistant Dean and Academic Coordinator for the College of Arts and Sciences at Washington University in St Louis

Introductory Chemistry Charles H. Corwin, 2011 To help students learn chemical skills and concepts more effectively *Introductory Chemistry Concepts and Critical Thinking* Sixth Edition highlights the connection between key concepts and key problem solving skills through critical thinking Math and problem solving are covered early in the text Corwin builds your problem solving ability through innovative learning aids and technology formulated to meet your needs This revision retains all the strengths of the previous editions while adding emphasis on conceptual understanding and critical thinking

Knowledge Spaces Jean-Claude Falmagne, Dietrich Albert, Christopher Doble, David Eppstein, Xiangen Hu, 2013-06-20 The book describes up to date applications and relevant theoretical results These applications come from various places but the most important one numerically speaking is the internet based educational system ALEKS The ALEKS system is bilingual English Spanish and covers all of mathematics from third grade to the end of high school and chemistry It is also widely used in higher education because US students are often poorly prepared when they reach the university level The chapter by Taagepera and Arasasingham deals with the application of knowledge spaces independent of ALEKS to the teaching of college chemistry The four chapters by Albert and his collaborators strive to give cognitive interpretations to the combinatoric structures obtained and used by the ALEKS system The contribution by Eppstein is technical and develops means of searching the knowledge structure efficiently

Survival Handbook for the New Chemistry Instructor Diane M. Bunce, Cinzia M. Muzzi, 2004 This book provides an overview of the issues facing new chemistry faculty in preparation for teaching Serving as a reference to answer specific questions new chemistry faculty encounter this book is comparable to sitting down with a colleague in the department and talking through some ideas or gaining some pointers on how to avoid common pitfalls It is the one single place new chemistry faculty can go to find practical information on how to teach and how to prepare for teaching their first course Chapters are written both by established experts in the field and by new professors within their first couple of years of teaching

General Chemistry Ralph H. Petrucci, William S. Harwood, 1993 *General Chemistry Principles and Modern Applications* is recognized for its superior problems lucid writing and precision of argument This updated and expanded edition retains the popular and innovative features of previous editions including Feature Problems follow up Integrative and Practice Exercises to accompany every in chapter Example and Focus On application boxes as well as new Keep in Mind marginal notes Topics covered include atoms and the atomic theory chemical compounds and reactions gases Thermochemistry electrons in atoms chemical bonding liquids solids and intermolecular forces chemical kinetics principles of

chemical equilibrium acids and bases electrochemistry representative and transitional elements and nuclear and organic chemistry For individuals interested in a broad overview of chemical principles and applications Organic Chemistry T. W. Graham Solomons, Craig B. Fryhle, Scott A. Snyder, 2022-01-19 Organic Chemistry 13th Edition continues Solomons Fryle and Snyder s tradition of excellence in teaching and preparing students for success in both the classroom and beyond Central to the authors is their approach in emphasizing organic chemistry s relationship between structure and reactivity To accomplish this the content is organized in a way that combines the most useful features of a functional group approach with one largely based on reaction mechanisms The authors philosophy is to emphasize mechanisms and their common aspects as often as possible and at the same time use the unifying features of functional groups as the basis for most chapters The structural aspects of the authors approach show students what organic chemistry is Mechanistic aspects of their approach show students how it works And wherever an opportunity arises the authors show students what it does in living systems and the physical world around us

ACTUAL ISSUES OF THE DEVELOPMENT OF SCIENCE AND ENSURING THE QUALITY OF EDUCATION, 2023-04-28 Proceedings of the XII International Scientific and Practical Conference **Unplugging the Classroom**

Hilary Anne Wilder, Sharmila Pixy Ferris, 2017-05-17 Unplugging the Classroom Teaching with Technologies to Promote Students Lifelong Learning provides techniques to help teaching and learning in an age where technology untethers instruction from the classroom from semester seat time and from a single source of expertise The book brings together researchers and practitioners from diverse academic fields including library perspectives and presents interdisciplinary discussions from both theoretical and applied areas It is unique in its goal of bringing educators and librarians together to explore the challenges that are faced by students and faculty in any time any place any path and any pace learning In spite of the fact that the mobile revolution has definitively arrived students and faculty alike aren t ready to make the leap to mobile learning The pressures of technological advances along with the changing nature of learning will demand increasingly profound changes in education Researchers have begun to address this issue but the revolution in mobile communication has not been accompanied by a concomitant growth in pedagogical resources for educators and students More importantly such growth needs to be under girded by sound learning theories and examples of best practice Provides a hands on resource useful to both novices and experts for technology enabled teaching and learning Gives both discipline specific and cross disciplinary perspectives Discusses discipline specific mobile applications Offers an opportunity to meet the needs of contemporary learners and foster their competencies as lifelong learners Addresses emerging issues in technology and pedagogy Mastering American English Rebecca E. Hayden, 1956

Cases on Critical and Qualitative Perspectives in Online Higher Education Orleans, Myron, 2014-01-31 Online education continues to permeate mainstream teaching techniques in higher education settings Teaching upper level classes in an online setting is having a major impact on education as a whole and is fundamentally altering global learning Cases on Critical and Qualitative Perspectives in Online

Higher Education offers a collection of informal personalized articles that identify describe and examine actual experiential domains of online program and course production Administrators developers instructors staff technical support and students in the field of online higher education will benefit from these case studies to reinforce and enhance their work

Succeeding in Organic Chemistry Joseph C. Sloop, 2010 This text is specifically designed to help introductory Organic Chemistry students Understand The fundamental concepts covered in undergraduate organic chemistry The purpose of this book is three fold To explode the misconceptions and misgivings that are prevalent regarding this vast subject provide additional insight for students on a number of concepts essential to mastery of organic chemistry and explore alternative learning strategies to assist the beginning organic chemistry student in applying a specialized problem solving technique which centers on structure function and a mechanistic approach Examples of key chemical transformations are dissected and analyzed to assist students in improving their problem solving skills Each chapter contains a number of additional problems And The solutions to those problems are provided at the end of each chapter How I Kicked The MCAT's Ass: Your Guide to Mastering Any Test , End-User Considerations in Educational Technology Design Roscoe, Rod D., Craig, Scotty D., Douglas, Ian, 2017-06-16 Emerging technologies have enhanced the learning capabilities and opportunities in modern school systems To continue the effective development of such innovations the intended users must be taken into account End User Considerations in Educational Technology Design is a pivotal reference source for the latest scholarly material on usability testing techniques and user centered design methodologies in the development of technological tools for learning environments Highlighting a range of pertinent topics such as multimedia learning human computer interaction and online learning this book is ideally designed for academics researchers school administrators professionals and practitioners interested in the design of optimized educational technologies **15 PGT Chemistry Test Papers EMRS** Mocktime Publication, EMRS Exam Teachers PGT Chemistry Test Papers 15 Practice Papers Tier 1 Eklaya Model Residential Schools as per Official Exam Pattern and Syllabus **Student Guide For Living Chemistry** Barbara Ucko, 2012-12-02 Student Guide for Living Chemistry is a 23 chapter textbook guide that allows students to study and review on their own and test their understanding to help them prepare for examinations Every chapter begins with a list of objectives stating exactly the skills to develop in a particular unit Each objective corresponds to a section in the textbook Living Chemistry Three kinds of questions are provided for each objective to check the student s understanding namely short answer Study Questions multiple choice and fill in The answers for all questions are provided at the end of the chapter The opening chapters cover the SI units composition of matter chemical bonding compounds chemical change gases respiration and water The subsequent chapters deal with solutions acids bases salts nuclear and organic chemistry oxygen derivatives and hydrocarbons polymers and other organic derivatives This textbook also explores the chemistry of carbohydrates lipids proteins enzymes and energy and carbohydrate metabolism The remaining chapters discuss the chemistry of vitamins

hormones body fluid drugs and poisons Undergraduate chemistry students will find this book invaluable *Fundamentals of Chemistry* Ralph A. Burns, 2003 For one semester preparatory chemistry courses or general purpose introductory chemistry courses This clearly written well illustrated versatile textbook provides thorough coverage of chemistry with a balance of problem solving skills real world applications and an emphasis on critical thinking and the process of science A supporting theme throughout the text continually emphasizes that chemistry is everywhere

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mole moles of H_2O 40.1822 moles therefore according to the equation

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time the total pressure $p_a \times \text{mole fraction}_a = \text{mole}_a / \text{moles total}$ right sum $n_1 \text{ all } x_{n_1}$ mole fraction of substance
a is often denoted by x_a and

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web 1 in the given mixture the sum of all the mole fractions is equal to one $x_a + x_b + \dots$ the multiplication of the mole fraction
by 100 gives the mole percentage 3 mole fraction is a unitless and dimensionless expression advantages of molar fraction
mole fraction is not dependent on the temperature

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web nov 6 2023 $x_a \text{ mol a} / \text{mol a} + \text{mol b} \text{ and } x_b \text{ mol b} / \text{mol a} + \text{mol b}$ note in the given mixture the sum of all the mole fractions
present is equal to one $x_a + x_b + \dots = 1$ when the mole fractions are multiplied by 100 they give the mole percentage mole fraction is
a unitless and dimensionless expression image will be uploaded soon

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ways to represent the concentration of the solution it is defined as the number of molecules of component a divided by the
total number of molecules of all the components in the solution

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web solution total mass of the answer 100 g mass of benzene 30 g therefore mass of carbon tet 100 - 30 g benzene s molar
mass C_6H_6 6 12 6 1 g mol 1 78 g mol 1 hence C_6H_6 s number of moles 30 / 78 mol 0.3846 mol molar mass of carbon tet CCl_4 1
12 4 355 154 g mol 1

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g of water calculate the molarity and molality of the solution given n 14 h 1 c 12 o 16 given mass of solute urea 11 11 g mass
of solvent water 100 g 0.1 kg to find molarity of solution solution

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of CH_3OH 0.1722017 mole fraction of CH_3OH 0.073 to solve more examples on mole fraction formula and practice more

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web you may work out the mole fractions on your own problem 5 what is the mass of a sample of a 0.449 molal KBr that contains 2.92 kg of water 1 molality moles solute divided by kilograms solute $0.449 \text{ mol/kg} \times 2.92 \text{ kg} \times 131.108 \text{ mol of KBr} \times 2 \text{ moles times molar mass equals grams}$ 131.108 mol 119.0023 g mol 156 g KBr

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web notice that the mole fraction has no units on it and is written as a decimal value do not change it to percent note of caution you could see the term mole percent it is simply the mole fraction multiplied by 100 for example in the problem just below the mole fraction of cinnamic acid is 0.2885 its mole percent would be 28.85

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web since the mole fraction of the O_2 is 0.050 it contributes 0.050 or if you will 5.0 of the total pressure thus total pressure is 0.2105042 atm problem 9 a sample of 1.43 g of helium and an unweighed quantity of O_2 are mixed in a flask at room temperature