

Name _____

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Reading the Periodic Table **KEY**

1. List the atomic numbers of the elements in Period 2.

3, 4, 5, 6, 7, 8, 9, 10

2. List the symbols of the elements in Period 3.

Na, Mg, Al, Si, P, S, Cl, Ar

3. Name the elements in Period 2 that are metals.

Li, Be

4. Name the elements in Period 1 that are nonmetals.

H, He

5. Which element in Period 3 has the least mass?

Sodium (Na)

6. Which element in Period 4 has the most mass?

Krypton (Kr)

7. To which group/family do these belong?

a. Cobalt **transition metals**

b. Sodium **alkali metals**

c. Argon **noble gases**

d. Silicon **group 14 metalloids**

e. Chlorine **halogens**

f. Magnesium **alkaline earth metals**

8. Find iron on the periodic table:

a. What is the atomic number of iron? **26**

b. To which group does iron belong? **Transition metals**

9. Name the elements that have properties similar to that of chlorine.

Fluorine, bromine, iodine, astatine, tennessine

10. Fill in the chart below. The first row has been completed as an example.

Element	Symbol	Atomic Number	Period	Group
sodium	Na	11	3	1
oxygen	O	8	2	16
krypton	Kr	36	4	18
gold	Au	79	6	11
cadmium	Cd	48	5	12
manganese	Mn	25	4	7

Answer Key To The Periodic Table Homework

Arshad Iqbal



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Table of Contents Answer Key To The Periodic Table Homework

1. Understanding the eBook Answer Key To The Periodic Table Homework
 - The Rise of Digital Reading Answer Key To The Periodic Table Homework
 - Advantages of eBooks Over Traditional Books
2. Identifying Answer Key To The Periodic Table Homework
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answer Key To The Periodic Table Homework
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answer Key To The Periodic Table Homework
 - Personalized Recommendations

- Answer Key To The Periodic Table Homework User Reviews and Ratings
- Answer Key To The Periodic Table Homework and Bestseller Lists
- 5. Accessing Answer Key To The Periodic Table Homework Free and Paid eBooks
 - Answer Key To The Periodic Table Homework Public Domain eBooks
 - Answer Key To The Periodic Table Homework eBook Subscription Services
 - Answer Key To The Periodic Table Homework Budget-Friendly Options
- 6. Navigating Answer Key To The Periodic Table Homework eBook Formats
 - ePub, PDF, MOBI, and More
 - Answer Key To The Periodic Table Homework Compatibility with Devices
 - Answer Key To The Periodic Table Homework Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answer Key To The Periodic Table Homework
 - Highlighting and Note-Taking Answer Key To The Periodic Table Homework
 - Interactive Elements Answer Key To The Periodic Table Homework
- 8. Staying Engaged with Answer Key To The Periodic Table Homework
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answer Key To The Periodic Table Homework
- 9. Balancing eBooks and Physical Books Answer Key To The Periodic Table Homework
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answer Key To The Periodic Table Homework
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Answer Key To The Periodic Table Homework
 - Setting Reading Goals Answer Key To The Periodic Table Homework
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Answer Key To The Periodic Table Homework
 - Fact-Checking eBook Content of Answer Key To The Periodic Table Homework

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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