



All answers should go inside the **RED BOXES**.

Name: _____

Date: _____

Student Exploration: Covalent Bonds

Vocabulary: covalent bond, diatomic molecule, Lewis diagram, molecule, noble gases, nonmetal, octet rule, shell, valence, valence electron

Prior Knowledge Questions (Do these **BEFORE** using the Gizmo.)

1. There are eight markers in a full set, but Flora and Frank each only have seven markers. Flora is missing the red marker, and Frank is missing the blue marker.

What can they do so that each has a full set of markers?

2. Otto and Olivia each have six markers. Otto is missing the purple and green markers, and Olivia is missing the black and brown markers. What can they do so that each has a full set?

Gizmo Warm-up

Just like the students described above, **nonmetal** atoms can share electrons. As you will see in the *Covalent Bonds* Gizmo, atoms form bonds in this way.

To begin, check that **Fluorine** is selected from the **Select a substance** menu. Click **Play** () to see the electrons orbiting the nucleus of each atom.



1. The outermost electrons in each atom are called **valence electrons**. How many valence electrons does each fluorine atom have?

2. Click **Pause** (). Drag an electron from the left atom to the right atom. Click **Play**. What happens?

3. Click **Pause**, drag an electron from the right atom to the left, and then click **Play**. What happens now?



Answers Of exploration Guide Covalent Bonds

Xianhai Huang, Robert G. Aslanian



Answers Of exploration Guide Covalent Bonds:

Computational Approaches for Chemistry Under Extreme Conditions Nir Goldman, 2019-02-18 This book presents recently developed computational approaches for the study of reactive materials under extreme physical and thermodynamic conditions. It delves into cutting edge developments in simulation methods for reactive materials including quantum calculations spanning nanometer length scales and picosecond timescales to reactive force fields, coarse grained approaches and machine learning methods spanning microns and nanoseconds and beyond. These methods are discussed in the context of a broad range of fields including prebiotic chemistry in impacting comets, studies of planetary interiors, high pressure synthesis of new compounds and detonations of energetic materials. The book presents a pedagogical approach for these state of the art approaches compiled into a single source for the first time. Ultimately the volume aims to make valuable research tools accessible to experimentalists and theoreticians alike for any number of scientific efforts spanning many different types of compounds and reactive conditions.

The Britannica Guide to Matter Erik Gregersen, Associate Editor, Astronomy and Space Exploration, 2011-01-15 The study of matter is the study of all material things as well as their ability to transform from one state to another. All matter assumes one of several basic states: solid, liquid, gas, and plasma, being the most common. Under varying conditions, each state can be altered to form new substances or adopt new characteristics. This insightful book covers the various structures and elements of different types of matter while examining the physical and chemical properties that allow for permutation and change.

Evaluation of Enzyme Inhibitors in Drug Discovery Robert A. Copeland, 2013-01-31 Offers essential guidance for discovering and optimizing novel drug therapies. Using detailed examples, *Evaluation of Enzyme Inhibitors in Drug Discovery* equips researchers with the tools needed to apply the science of enzymology and biochemistry to the discovery, optimization, and preclinical development of drugs that work by inhibiting specific enzyme targets. Readers will applaud this book for its clear and practical presentations, including its expert advice on best practices to follow and pitfalls to avoid. This Second Edition brings the book thoroughly up to date with the latest research findings and practices. Updates explore additional forms of enzyme inhibition and special treatments for enzymes that act on macromolecular substrates. Readers will also find new discussions detailing the development and application of the concept of drug target residence time. *Evaluation of Enzyme Inhibitors in Drug Discovery* begins by explaining why enzymes are such important drug targets and then examines enzyme reaction mechanisms. The book covers reversible modes of inhibitor interactions with enzymes, assay considerations for compound library screening, lead optimization and structure activity relationships for reversible inhibitors, slow binding and tight binding inhibitors, drug target residence time, irreversible enzyme inactivators. The book ends with a new chapter exploring the application of quantitative biochemical principles to the pharmacologic evaluation of drug candidates during lead optimization and preclinical development. The Second Edition of *Evaluation of Enzyme Inhibitors in Drug Discovery* continues to offer a treatment of enzymology applied to

drug discovery that is quantitative and mathematically rigorous At the same time the clear and simple presentations demystify the complex science of enzymology making the book accessible to many fields from pharmacology to medicinal chemistry to biophysics to clinical medicine

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