



NAME: Abhishek H. Nair Div: A Roll No. 1

SCIENCE Paper – 2*(Two hours)**You will not be allowed to write during the first 15 minutes.**This time is to be spent in reading the Question Paper.**The time given at the head of this paper is the time allowed for writing the answers.**Section I is compulsory. Attempt any four questions from Section II.**The intended marks for questions or parts of questions are given in brackets []***SECTION I (40 Marks)***(Attempt ALL questions)***Question 1****[15]**

Choose the correct answer from the options given below:

(Do not copy the question; write the correct answers only.)

- Whenever the gases react under similar conditions of temperature and pressure, they do so in volumes which bear a simple ratio of each other and the products if gaseous. The above law was stated by
 a) Boyle b) Charles c) Gay Lussac d) Avogadro
- A strip of copper is placed in four different colourless salt solutions. After electrolysis, one of the salt solutions becomes blue in colour. Identify the salt solution which turns blue from the given salt solutions:
 a) K_2SO_4 b) $CuSO_4$ c) $AgSO_4$ d) $CaSO_4$
- Potassium is the element below sodium in the long form of the periodic table. The electronegativity of potassium will be
 a) Lower than sodium b) Higher than sodium c) Same as sodium d) None of these
- The salt solution which does not react with ammonium hydroxide is :
 a) Calcium Nitrate b) Zinc Nitrate c) Lead Nitrate d) Copper Nitrate
- Assertion :** Sodium is the most reactive metal among group 1 elements.
Reason : In case of metals, greater the tendency to lose electrons, greater is the reactivity.
 a) A and R both are correct statements and R is the correct explanation of the A
 b) A and R both are correct statements but R is not the correct explanation of the A
 c) A is true, but R is false
 d) A is false , but R is true



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Table of Contents Answers For A Ls Chemistry

1. Understanding the eBook Answers For A Ls Chemistry
 - The Rise of Digital Reading Answers For A Ls Chemistry
 - Advantages of eBooks Over Traditional Books
2. Identifying Answers For A Ls Chemistry
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answers For A Ls Chemistry
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answers For A Ls Chemistry
 - Personalized Recommendations
 - Answers For A Ls Chemistry User Reviews and Ratings
 - Answers For A Ls Chemistry and Bestseller Lists

5. Accessing Answers For A Ls Chemistry Free and Paid eBooks
 - Answers For A Ls Chemistry Public Domain eBooks
 - Answers For A Ls Chemistry eBook Subscription Services
 - Answers For A Ls Chemistry Budget-Friendly Options
6. Navigating Answers For A Ls Chemistry eBook Formats
 - ePub, PDF, MOBI, and More
 - Answers For A Ls Chemistry Compatibility with Devices
 - Answers For A Ls Chemistry Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answers For A Ls Chemistry
 - Highlighting and Note-Taking Answers For A Ls Chemistry
 - Interactive Elements Answers For A Ls Chemistry
8. Staying Engaged with Answers For A Ls Chemistry
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answers For A Ls Chemistry
9. Balancing eBooks and Physical Books Answers For A Ls Chemistry
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answers For A Ls Chemistry
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Answers For A Ls Chemistry
 - Setting Reading Goals Answers For A Ls Chemistry
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Answers For A Ls Chemistry
 - Fact-Checking eBook Content of Answers For A Ls Chemistry
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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