Chemistry Final Exam Review Questions

Useful Values and Equations

Some of the following values and equations may be useful for answering some of the questions and problems.

9P = 1.8(9C) + 32speed of light = 3.00×10^8 m/s $c = \lambda w$

R = 0.082057 L-atm / deg-mole Planck's constant = 6.63 × 10-34 J·s Fig. 1800

Multiple Choice

Identify the letter of the choice that best completes the statement or answers the question.

- One of the following does not describe gases. Which one is it?
 - a. Expands infinitely.

Easily compressed.

Rigid shape.

- d. Relatively low densities.
- The following properties describe the element, mercury (Hg). Which one is a chemical property?
 - a. Its density is 5.43 g/cm³.
 - b. It is a silvery-white metal.

- c. It is a liquid at 298 K.
- d. Mercury(II) sulfide forms when it is exposed to sulfide ions.
- All of the following are properties of SO₂. Which one is a physical property?
 - It reacts with oxygen to form SO₃².

- e. It is a strong reducing agent.
- mospheric pressure.
- It condenses to a colorless liquid at 14°F and at It reacts with water and oxygen to form an acid...
- 4. Which of the following numbers has 4 significant figures?

b. 0.0430

c. 0.0431

- 0.43980
- 5. How many significant figures should there be in the answer to the following math equation? 2.834 + 5.71520 + 2.12 + 178.1 + 250.2619
 - a. 3

e. 5

- 6. How many significant figures should there be in the answer to the following math equation? $23.413 \text{ g} + (2.15 \text{ cm} \times 1.1 \text{ cm} \times 3.73 \text{ cm})$

b. 3

- 7. At 25°C, one cm3 of iron has a mass of 7.62 grams. What volume of iron is occupied by 125 grams of iron?
 - a. 0.82 cm³

- b. 8.2 cm³
- c. 0.164 cm³

- d. 16.4 cm³
- 8. A cube of lead has a volume of 6.61 cm3. Its mass is 75.0 grams. What is its density?
 - a. 3.85 g/cm³
- b. 7.41 g/cm³
- c. 11.3 g/cm³
- d. 22.6 g/cm³

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Table of Contents Answers For Chemistry Final Exam Review 2014

- 1. Understanding the eBook Answers For Chemistry Final Exam Review 2014
 - The Rise of Digital Reading Answers For Chemistry Final Exam Review 2014
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Answers For Chemistry Final Exam Review 2014
 - Exploring Different Genres
 - o Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - o Features to Look for in an Answers For Chemistry Final Exam Review 2014
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Answers For Chemistry Final Exam Review 2014
 - Personalized Recommendations
 - Answers For Chemistry Final Exam Review 2014 User Reviews and Ratings
 - Answers For Chemistry Final Exam Review 2014 and Bestseller Lists

- 5. Accessing Answers For Chemistry Final Exam Review 2014 Free and Paid eBooks
 - Answers For Chemistry Final Exam Review 2014 Public Domain eBooks
 - Answers For Chemistry Final Exam Review 2014 eBook Subscription Services
 - Answers For Chemistry Final Exam Review 2014 Budget-Friendly Options
- 6. Navigating Answers For Chemistry Final Exam Review 2014 eBook Formats
 - o ePub, PDF, MOBI, and More
 - Answers For Chemistry Final Exam Review 2014 Compatibility with Devices
 - Answers For Chemistry Final Exam Review 2014 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answers For Chemistry Final Exam Review 2014
 - Highlighting and Note-Taking Answers For Chemistry Final Exam Review 2014
 - Interactive Elements Answers For Chemistry Final Exam Review 2014
- 8. Staying Engaged with Answers For Chemistry Final Exam Review 2014
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answers For Chemistry Final Exam Review 2014
- 9. Balancing eBooks and Physical Books Answers For Chemistry Final Exam Review 2014
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answers For Chemistry Final Exam Review 2014
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Answers For Chemistry Final Exam Review 2014
 - Setting Reading Goals Answers For Chemistry Final Exam Review 2014
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Answers For Chemistry Final Exam Review 2014
 - Fact-Checking eBook Content of Answers For Chemistry Final Exam Review 2014
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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