## **Chemistry Study Guide**



#### Objective:

To understand the structure of an atom, and the nature + properties of subatomic particles 🛫

#### Matter:

Anything that has mass and occupies space 💥 🕥

Examples: air, gases, liquids, solidz!

#### Atoms:

Atoms are the smallest, indivisible particle &.

- Outer part → shell/orbit/energy level → carries the electronz (e-)
- Inner part → nucleus → houses the protons (p+) and neutronz (n)
- Outer & Inner part → together these are the subatomic particles

#### Subatomic Particle Table:

| Particle | Symbol | Mass (amu) | Charge | Location    |
|----------|--------|------------|--------|-------------|
| Proton   | p."    | 1          | -1     | Nucleus     |
| Neutron  | m      | 1          | 0      | Nucleus     |
| Electron | •      | -0         | -1     | Orbit/Shell |

# **Answers For Holt Chemistry Study Guide**

**Yijin Wang** 

### **Answers For Holt Chemistry Study Guide:**

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