

CHAPTER 7 REVIEW

Chemical Formulas and Chemical Compounds**SECTION 1****SHORT ANSWER** Answer the following questions in the space provided.

1. _____ In a Stock system name such as iron(III) sulfate, the Roman numeral tells us
 - (a) how many atoms of Fe are in one formula unit.
 - (b) how many sulfate ions can be attached to the iron atom.
 - (c) the charge on each Fe ion.
 - (d) the total positive charge of the formula unit.
2. _____ Changing a subscript in a correctly written chemical formula
 - (a) changes the number of moles represented by the formula.
 - (b) changes the charges on the other ions in the compound.
 - (c) changes the formula so that it no longer represents the compound it previously represented.
 - (d) has no effect on the formula.
3. The explosive TNT has the molecular formula $C_7H_5(NO_2)_3$.
 - _____ a. How many elements make up this compound?
 - _____ b. How many oxygen atoms are present in one molecule of $C_7H_5(NO_2)_3$?
 - _____ c. How many atoms in total are present in one molecule of $C_7H_5(NO_2)_3$?
 - _____ d. How many atoms are present in a sample of 2.0×10^{23} molecules of $C_7H_5(NO_2)_3$?
4. How many atoms are present in each of these formula units?
 - _____ a. $Ca(HCO_3)_2$
 - _____ b. $C_{12}H_{22}O_{11}$
 - _____ c. $Fe(ClO_2)_3$
 - _____ d. $Fe(ClO_3)_2$
5. _____
 - _____ a. What is the formula for the compound dinitrogen pentoxide?
 - _____ b. What is the Stock system name for the compound FeO ?
 - _____ c. What is the formula for sulfurous acid?
 - _____ d. What is the name for the acid H_3PO_4 ?

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Kaplan Test Prep



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