

2017 AP Chemistry FRQ

Question 3



3. At high temperatures, $\text{N}_2(\text{g})$ and $\text{O}_2(\text{g})$ can react to produce nitrogen monoxide, $\text{NO}(\text{g})$, as represented by the equation above.
- (a) Write the expression for the equilibrium constant, K_p , for the forward reaction.
- (b) A student injects $\text{N}_2(\text{g})$ and $\text{O}_2(\text{g})$ into a previously evacuated, rigid vessel and raises the temperature of the vessel to 2000°C . At this temperature the initial partial pressures of $\text{N}_2(\text{g})$ and $\text{O}_2(\text{g})$ are 6.01 atm and 1.61 atm, respectively. The system is allowed to reach equilibrium. The partial pressure of $\text{NO}(\text{g})$ at equilibrium is 0.122 atm. Calculate the value of K_p .

2008 Chemistry Frq Answers

Kristin Tuttle Bump



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Compendium of Science Demonstration-related Research from 1918 to 2008 David Michael Majerich, Joseph S. Schmuckler, Kathleen Fadigan, 2008 This compendium highlights a ninety year history of science education focused on how demonstrations have been used for the express teaching and learning of science at the high school college and university levels It is presented in three chapters a Experimental Studies Comparing the Lecture Demonstration Method and the Individual Laboratory Method of Teaching Science 1918 1989 b Further Types of Demonstration related Comparison Studies 1958 2008 and c More Recent Demonstration related Non Comparative Studies 1980 2008 Organization of the research in this manner revealed that an interest in investigating this type of research by science educators had and continues to evolve This work was constructed from an intense and nearly exhaustive review of the research While the authors believe that it can be misconstrued that the scholarly literature and empirical research are one in the same we can show that it is necessary to make a distinction between the two sources of information Upon review of the scholarly literature and empirical research on the demonstration topics a salient outcome is that these two bodies of information are only loosely corroborated From this point on we use the terms literature and research when referring to each of these bodies of information To begin we describe studies comparing the effectiveness of the methods of lecture demonstration and the individual laboratory The very early attempts 1918 1964 to determine the superiority of the lecture demonstration method over the individual laboratory method of teaching science and vice versa remained unresolved through the mid 1960s Cunningham 1946 suggested several reasons for the inconclusive nature of the results reported by researchers of lecture demonstration and individual laboratory comparison studies After carefully reviewing the studies up to 1946 he noted that there were numerous variables that should have remained fixed for the duration of the experiments Some of the variables that confounded the results of those investigations included the uncontrolled variables related to a the teacher b the complexity of experiments and apparatus c the time spent on each method d the amount of science studied by students and e the performer of the demonstrations Even after exacting a procedure to control for the teacher variable

reported by Cunningham 1946 Yager et al 1969 were able to show that students who participated in a discussion demonstration group or discussion laboratory group developed more skills than those students who received science instruction via a discussion only method The results of Yager and his predecessors suggested that neither the demonstration method nor the laboratory method for the teaching and learning of science was superior From 1958 to 2008 investigations comprised of multiple demonstration related comparison studies were also unable to confirm the superiority of the demonstration method for the teaching and learning of science as compared to other methods in each study For instance Oliver 1975 compared three methods of teaching high school biology lecture discussion a combination of lecture discussion and demonstration and demonstrations Initially although he determined that the lecture discussion method of teaching biology was noticeably superior in terms of biology content acquisition during the onset of the semester this result was ephemeral and dissipated at the termination of the following semester Comparing a self paced instructional method and a teacher demonstration method of teaching college chemistry Eniaiyaju 1983 concluded that students achievement scores were higher when they participated in the self paced program in *Concepts of Matter in Science Education* Georgios Tsaparlis, Hannah Sevan, 2013-07-09 Bringing together a wide collection of ideas reviews analyses and new research on particulate and structural concepts of matter *Concepts of Matter in Science Education* informs practice from pre school through graduate school learning and teaching and aims to inspire progress in science education The expert contributors offer a range of reviews and critical analyses of related literature and in depth analysis of specific issues as well as new research Among the themes covered are learning progressions for teaching a particle model of matter the mental models of both students and teachers of the particulate nature of matter educational technology chemical reactions and chemical phenomena chemical structure and bonding quantum chemistry and the history and philosophy of science relating to the particulate nature of matter The book will benefit a wide audience including classroom practitioners and student teachers at every educational level teacher educators and researchers in science education If gaining the precise meaning in particulate terms of what is solid what is liquid and that air is a gas were that simple we would not be confronted with another book which while suggesting new approaches to teaching these topics confirms they are still very difficult for students to learn Peter Fensham Emeritus Professor Monash University Adjunct Professor QUT from the foreword to this book **Chemistry** Rob Ritchie, 2008-07 Divided into seven manageable day sections this timed revision programme covers essential GCSE topics in double page spreads These spreads indicate how much time should be spent on each section and combine clear and concise explanations flow charts spidergrams and illustrations with progress check questions and answers *AP Chemistry For Dummies* Peter J. Mikulecky, Michelle Rose Gilman, Kate Brutlag, 2008-11-13 A practical and hands on guide for learning the practical science of AP chemistry and preparing for the AP chem exam Gearing up for the AP Chemistry exam *AP Chemistry For Dummies* is packed with all the resources and help you need to do your very best Focused on the chemistry

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be trained to meet the demands and opportunities afforded by this technology rich landscape The Handbook of Research on Teacher Education in the Digital Age focuses on the needs of teachers as they redesign their curricula and lessons to incorporate new technological tools Including theoretical frameworks empirical research and best practices this book serves as a guide for researchers educators and faculty and professional developers of distance learning tools **Intelligent Tutoring Systems** Beverly Woolf, Esma Aimeur, Roger Nkambou, Susanne Lajoie, 2008-07-08 This book constitutes the refereed proceedings of the 9th International Conference on Intelligent Tutoring Systems ITS 2008 held in Montreal Canada in June 2008 The 63 revised full papers and 61 poster papers presented together with abstracts of 5 keynote talks were carefully reviewed and selected from 207 submissions The papers are organized in topical sections on emotion and affect tutor evaluation student modeling machine learning authoring tools tutor feedback and intervention data mining e learning and Web based ITS natural language techniques and dialogue narrative tutors and games semantic Web and ontology cognitive models and collaboration

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