AP Chemistry Chapter 4 Practice Exam

- The Claus reactions, shown below, are used to generate elemental sulfur from hydrogen sulfide.

$$2H_2S + 3O_2 \rightarrow 2SO_2 + 2H_2O$$

 $SO_2 + 2H_2S \rightarrow 3S + 2H_2O$

How much sulfur (in grams) is produced from 21.5 grams of O₂? 21.5502 |mol 02 | 2503 | 35 | 32.0675 = 43.155

- A) 97.0 g
- B) 14.4 g
- C) 21.5 g
- none of these
- B
- Iron is produced from its ore by the reactions:

$$2C(e) + O_2(e) \rightarrow 2CO(e)$$

$$Fe_2O_3(s) + 3CO(g) \rightarrow 2Fe(s) + 3CO_3(g)$$

Closes:

How many moles of $O_2(g)$ are needed to produce 4.6 moles of Fe(s)?

- A) 2.3 mole O₂ B) 3.5 mole O₂
- C) 4.6 mole O₂
- D) 6.9 mole O₂
- E) 13.8 mole O₂

- B
- Suppose the reaction Ca₂(PO₄)₂ + 3H₂SO₄ → 3CaSO₄ + 2H₂PO₄ is carried out starting with 153 g. of $Ca_2(PO_k)_2$

and 76.8 g of H_2SO_4 . How much phosphoric acid will be produced?

A) 76.7 gB) 51.1 g1 $C_{0.3}(C_{0.3}) = C_{0.3}(C_{0.3}) = C_{0.3}(C$

- B) 51.1 g C) 229.8 g

 - D) 115.1 c E) 96.7 g
- 7 6.89 H2504 | Ind H2504 2 H2 POY 1 3 =

1

- - SO₂ reacts with H₂S as follows:

$$2H_2S + SO_2 \rightarrow 3S + 2H_2O$$

When 7.50 g of H₂S reacts with 12.75 g of SO₂, which statement applies?

- A) 6.38 g of sulfur are formed.
- B) 10.6 g of sulfur are formed.
- C) 0.0216 moles of H₂S remain.
- D) 1.13 g of H₂S remain.
- E) SO₂ is the limiting reagent.

4.6 mil Fersi 300 102 = 3.45mol02

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Norman Reid

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education It is hoped that this book will prove helpful to all involved in sharing the exciting story of the way humans have come to understand the molecular world one of the great examples of great human endeavour

The Science Teacher, 2009

Structural Biology in Drug Discovery Jean-Paul Renaud,2020-02-05 With the most comprehensive and up to date overview of structure based drug discovery covering both experimental and computational approaches Structural Biology in Drug Discovery Methods Techniques and Practices describes principles methods applications and emerging paradigms of structural biology as a tool for more efficient drug development Coverage includes successful examples academic and industry insights novel concepts and advances in a rapidly evolving field The combined chapters by authors writing from the frontlines of structural biology and drug discovery give readers a valuable reference and resource that Presents the benefits limitations and potentiality of major techniques in the field such as X ray crystallography NMR neutron crystallography cryo EM mass spectrometry and other biophysical techniques and computational structural biology Includes detailed chapters on druggability allostery complementary use of thermodynamic and kinetic information and powerful approaches such as structural chemogenomics and fragment based drug design Emphasizes the need for the in depth biophysical characterization of protein targets as well as of therapeutic proteins and for a thorough quality assessment of experimental structures Illustrates advances in the field of established therapeutic targets like kinases serine proteinases GPCRs and epigenetic proteins and of more challenging ones like protein protein interactions and intrinsically disordered proteins

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