

PHYSICS

QUESTIONS

1. According to the Bohr theory, which of the following transitions will give rise to the most energetic photon?

- (1) $n=3 \rightarrow n=2$ (2) $n=4 \rightarrow n=3$
(3) $n=4 \rightarrow n=2$ (4) $n=3 \rightarrow n=1$

2. A car has a mass of 1500 kg and speed of 20 m/s. If the speed has to be measured with the accuracy of 1%, then the uncertainty in its position is

- (1) 1.0×10^{-16} m (2) 1.0×10^{-17} m
(3) 1.0×10^{-18} m (4) 1.0×10^{-19} m

3. The ground state of hydrogen atom is at the ground state is 13.6 eV. The value of the first excited state of hydrogen atom is

- (1) 13.6 eV (2) 3.4 eV
(3) 10.2 eV (4) 20.4 eV

4. The sample of a pure, first excited state of hydrogen atom is released during ionization of hydrogen atom. The number of photons emitted per second is

- (1) 0 (2) 1
(3) 2 (4) 3

5. The mass of a proton is 1.67×10^{-27} kg. The mass of a neutron is 1.67×10^{-27} kg. The mass of a deuteron is 3.34×10^{-27} kg. The mass of a triton is 5.01×10^{-27} kg.

- (1) 1.67×10^{-27} kg (2) 3.34×10^{-27} kg
(3) 5.01×10^{-27} kg (4) 6.68×10^{-27} kg

6. The value of $\left(\frac{dE}{d\lambda}\right)$ is maximum for

- (1) $\lambda = 0$ (2) $\lambda = \infty$
(3) $\lambda = \frac{h}{mv}$ (4) $\lambda = \frac{h}{mv} + \frac{h}{mv}$

7. Which of the following set of quantum numbers is not possible?

- (1) $n=1, l=0, m=0$
(2) $n=2, l=1, m=1$
(3) $n=3, l=2, m=2$
(4) $n=4, l=3, m=3$

8. Maximum kinetic energy of photoelectrons emitted when a light of frequency 1.5×10^{15} Hz is incident on a metal surface whose threshold frequency is equal to 1.0×10^{15} Hz, is 1.0×10^{-19} J.

- (1) 1.0×10^{-19} J (2) 1.0×10^{-18} J
(3) 1.0×10^{-17} J (4) 1.0×10^{-16} J

9. The number of photons of light having wavelength 660 nm which can provide 100 J of energy is about

- (1) 1.2×10^{21} (2) 1.2×10^{22}
(3) 1.2×10^{23} (4) 1.2×10^{24}

10. Which of the following arrangement shows the correct order of maximum multiplicity?

- (1) $\uparrow \uparrow \uparrow \uparrow \uparrow \uparrow$ (2) $\uparrow \uparrow \uparrow \uparrow \uparrow \uparrow$
(3) $\uparrow \uparrow \uparrow \uparrow \uparrow \uparrow$ (4) $\uparrow \uparrow \uparrow \uparrow \uparrow \uparrow$

11. Which of the following arrangement shows the correct order of maximum multiplicity?

- (1) $\uparrow \uparrow \uparrow \uparrow \uparrow \uparrow$ (2) $\uparrow \uparrow \uparrow \uparrow \uparrow \uparrow$
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12. Which of the following arrangement shows the correct order of maximum multiplicity?

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14. Which of the following arrangement shows the correct order of maximum multiplicity?

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(3) $\uparrow \uparrow \uparrow \uparrow \uparrow \uparrow$ (4) $\uparrow \uparrow \uparrow \uparrow \uparrow \uparrow$

Space for Rough Work

PHYSICS

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1. Which of the following set of orbitals will have maximum density along the z-axis?

- (1) d_{xy}, d_{yz}

- (2) d_{yz}, d_{zx}

- (3) d_{zx}, d_{xy}

- (4) d_{xy}, d_{yz}

2. Maximum wavelength of spectral series of Balmer series for hydrogen atom is λ_0 . The following statement is

- (1) $\frac{1}{\lambda_0} = \frac{1}{\lambda_0}$

- (2) $\frac{1}{\lambda_0} = \frac{1}{\lambda_0}$

- (3) $\frac{1}{\lambda_0} = \frac{1}{\lambda_0}$

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[BIOLOGY]

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AN Whitehead



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