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CHEMISTRY LAB MANUAL

Class
XII

A Complete Lab Activity Book



SALIENT FEATURES

- Lab Experiments
- Investigatory Project
- Viva Voce



Aakash Chemistry Lab Manual

**Who's Who Marquis, Marquis Who's
Who**



Aakash Chemistry Lab Manual:

Quantitative Analysis Laboratory Manual Dr. Bupinderpal Singh ,Dr. Bikramjit Singh,Mr. Akash Bans,2025-06-29

Energetic Materials Mark J. Mezger,Kay J. Tindle,Michelle Pantoya,Lori J. Groven,Dilhan Kalyon,2017-07-06 This book will take an in depth look at the technologies processes and capabilities to develop and produce next generation energetic materials for both commercial and defense applications including military mining operations oil production and well perforation and construction demolition It will serve to highlight the critical technologies latest developments and the current capability gaps that serve as barriers to military fielding or transition to the commercial marketplace It will also explain how the processing technologies can be spun out for use in other non energetics related industries **Textile**

Chemist and Colorist ,1995 **Who's Who in Science and Engineering 2008-2009** Who's Who Marquis,Marquis Who's Who,2007-12 *Books in Print Supplement* ,2002 Who's who in the West ,1998 **Laboratory Manual for Chemistry**

,1966 *General Chemistry Lab Manual* Helen Mango,2014 **Practical/Laboratory Manual Chemistry Class - XI** Er. Meera Goyal,2021-05-29 1 Basic Laboratory Techniques 1 To cut a glass tube or glass rod 2 To bend the glass rod at an angle 3 To draw a glass jet from a glass tube 4 To bore a cork and fit a glass tube into it Viva Voce 2 Characterisation and Purification of Chemical Substances 1 To determine the melting point of the given unknown organic compound and its identification simple laboratory technique Viva Voce 2 To determine the boiling point of a given liquid when available in small quantity simple laboratory method Viva Voce 3 To prepare crystals of pure potash alum $K_2SO_4 \cdot Al_2(SO_4)_3 \cdot 24H_2O$ from the given impure sample 4 To prepare the pure crystals of copper sulphate from the given crude sample 5 To prepare pure crystals of benzoic acid from a given impure sample Viva Voce 3 Measurement of pH Values 1 To determine the pH value of vegetable juices fruit juices tap water and washing soda by using universal pH paper 2 To determine and compare the pH values of solutions of strong acid HCl and weak acid CH_3COOH of same concentration 3 To study the pH change in the titration of strong base Vs strong acid by using universal indicator paper 4 To study the pH change by common ion CH_3COO^- ion in case of weak acid CH_3COOH 5 To determine the change in pH value of weak base NH_4OH in presence of a common ion NH_4^+ Viva Voce 4 Chemical Equilibrium 1 To study the shift in equilibrium between ferric ions and thiocyanate ions by changing the concentrations of either of the ions 2 To study the shift in equilibrium between $Co(H_2O)_6^{2+}$ and Cl^- ions by changing the concentrations of either of the ions Viva Voce 5 Quantitative Analysis 1 To prepare M 10 oxalic acid solution by direct weighing method 2 To prepare M 10 solution of sodium carbonate by direct weighing method 3 To determine the strength of given solution of sodium hydroxide by titrating it against N 10 or M 20 solution of oxalic acid 4 To determine the strength of a given solution of hydrochloric acid by titrating it against a standard N 10 or M 20 sodium carbonate solution Viva Voce 6 Qualitative Analysis Analysis of Anions Analysis of Cations Viva Voce 7 Detection of Elements in Organic Compounds 1 To detect the presence of nitrogen sulphur and halogens in a given organic compound by Lassaigne s test 2 To

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Chemistry Lab Manual Neena Sinha, R Rangarajan, R P Manchanda, R K Gupta, Rajesh Kumar, Lab Manual

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