

Yes, "alkaline" earth metals produce alkaline hydroxides upon reaction with water.

The pink color of the phenolphthalein indicates this.

4. Compare the reactions of Ca with those of Mg. Which metal reacted more vigorously with water? Use observations from the lab activity to support your answer. Find strontium (Sr) on the periodic table and predict whether strontium is more or less reactive than calcium.

Calcium reacted much more vigorously, as indicated by heat production and copious quantities of calcium hydroxide. Magnesium reacted only slightly with water. Strontium is below calcium on the periodic table. The trend suggests that strontium should react even more vigorously than calcium.

5. Was there any evidence that Al reacted with water? What metal is used to make soft drink cans? Is aluminum more or less reactive than Alkaline Earth metals?

There was no evidence of reaction between aluminum and water. Soft drink cans are made of aluminum.

It is possible to use aluminum for drink cans because aluminum is significantly less reactive than

Alkaline Earth metals.

Alkali Metals and their reactions

Watch the video and record observations for each metal in Data Table 3. What do you see happen?

[Appearance of Pure Alkali Metals](#) [Reactions of Alkali Metals in Water and Products](#)

[Reactions of Alkali Metals in water](#) [A few more](#)

Data Table 3: [observations should be recorded during the videos](#)

	lithium	sodium	potassium	rubidium	cesium
Appearance of metal					
Reaction with water					
Products of reaction with water					

Answer each question using complete thoughts. Do not use the words "it" or "they" in your answers.

Alkaline Earth Metals Lab Answer Key

K Morrison



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