



Analytical Chemistry Interview Questions And Answers Guide

- Q1:** What is the difference between qualitative and quantitative analysis?
- A1:** Qualitative analysis is the determination of the presence or absence of a substance in a sample, while quantitative analysis is the determination of the amount of a substance in a sample.
- Q2:** What is the difference between a standard solution and a sample solution?
- A2:** A standard solution is a solution of known concentration, while a sample solution is a solution of unknown concentration.
- Q3:** What is the difference between a primary standard and a secondary standard?
- A3:** A primary standard is a substance that is pure, stable, and has a high molecular weight, while a secondary standard is a substance that is not pure, stable, or has a low molecular weight.
- Q4:** What is the difference between a titration and a gravimetric analysis?
- A4:** A titration is a method of determining the concentration of a substance in a solution by reacting it with a known volume of a standard solution, while gravimetric analysis is a method of determining the concentration of a substance in a sample by weighing it.
- Q5:** What is the difference between a colorimetric method and a spectrophotometric method?
- A5:** A colorimetric method is a method of determining the concentration of a substance in a solution by measuring the color of the solution, while a spectrophotometric method is a method of determining the concentration of a substance in a solution by measuring the absorbance of the solution.
- Q6:** What is the difference between a chromatographic method and a mass spectrometric method?
- A6:** A chromatographic method is a method of separating and identifying the components of a mixture, while a mass spectrometric method is a method of determining the mass of a substance in a sample.
- Q7:** What is the difference between a gas chromatographic method and a liquid chromatographic method?
- A7:** A gas chromatographic method is a method of separating and identifying the components of a mixture in the gas phase, while a liquid chromatographic method is a method of separating and identifying the components of a mixture in the liquid phase.
- Q8:** What is the difference between a gas-liquid chromatographic method and a gas-solid chromatographic method?
- A8:** A gas-liquid chromatographic method is a method of separating and identifying the components of a mixture in the gas phase using a liquid stationary phase, while a gas-solid chromatographic method is a method of separating and identifying the components of a mixture in the gas phase using a solid stationary phase.
- Q9:** What is the difference between a gas-liquid chromatographic method and a gas-solid chromatographic method?
- A9:** A gas-liquid chromatographic method is a method of separating and identifying the components of a mixture in the gas phase using a liquid stationary phase, while a gas-solid chromatographic method is a method of separating and identifying the components of a mixture in the gas phase using a solid stationary phase.
- Q10:** What is the difference between a gas-liquid chromatographic method and a gas-solid chromatographic method?
- A10:** A gas-liquid chromatographic method is a method of separating and identifying the components of a mixture in the gas phase using a liquid stationary phase, while a gas-solid chromatographic method is a method of separating and identifying the components of a mixture in the gas phase using a solid stationary phase.

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