



Name: \_\_\_\_\_ Date: \_\_\_\_\_

## Student Exploration: Nuclear Decay

**Background:** Alpha particles, beta particles, gamma rays, and neutrons are particles that are emitted from unstable nuclei. Alpha particles consist of two protons and two neutrons, beta particles consist of one electron or one positron, gamma rays consist of high-energy photons, and neutrons consist of one neutron.

**Prior Knowledge Questions:** (Write what you know about the following.)

1. What are some properties of atoms, isotopes, and radioactive materials? (What are **isotopes**? **radioactive**? How do isotopes of an element differ? What is an **isotope**?)

| Particle | Composition              | Charge   | Approximate mass |
|----------|--------------------------|----------|------------------|
| Alpha    | 2 protons, 2 neutrons    | +2       | 4 u              |
| Beta     | 1 electron or 1 positron | -1 or +1 | 1/1836 u         |
| Gamma    | High-energy photon       | 0        | 0                |

1. The **atomic number** of an atom is equal to the total number of protons in the nucleus.

Alpha particles have 2 protons, and  $\beta^-$  particles have 1 electron. What is the total number of the particle? **2**

2. The **mass number** of an atom is equal to the total number of protons and neutrons in the nucleus. All isotopes of an element have the same number of protons, but a different number of neutrons. All isotopes of an element have the same **atomic number**, but a different **mass number**.

### Atomic Structure

Atoms consist of protons, neutrons, and electrons. Protons and neutrons are in the nucleus, and electrons are in the electron cloud.

**Radioactivity:** The process by which an unstable nucleus loses energy by emitting ionizing particles and gamma rays.

There are three types of nuclear decay: alpha decay, beta decay, and gamma decay. Alpha decay involves the emission of an alpha particle, beta decay involves the emission of a beta particle, and gamma decay involves the emission of a gamma ray.



1. Alpha decay ( $\alpha$ ), beta decay ( $\beta^-$ ), and gamma decay ( $\gamma$ ) are the three types of nuclear decay. Alpha decay involves the emission of an alpha particle, beta decay involves the emission of a beta particle, and gamma decay involves the emission of a gamma ray.

2. Alpha decay involves the emission of an alpha particle, which is a helium nucleus. Beta decay involves the emission of a beta particle, which is an electron or a positron. Gamma decay involves the emission of a gamma ray, which is a high-energy photon.

**Radioactivity** is the process by which an unstable nucleus loses energy by emitting ionizing particles and gamma rays.



# Answer Sheet To Student Exploration Nuclear Decay In Explore Learning

**Padhraic Smyth**



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