Mastering Biology ch 2,3,4,5 study aid

Ch

2 - answer

A substance that cannot be broken down into other substances by ordinary chemical procedures is a(n) ______ . - answer element

Which of the following is a trace element required by most living organisms? - answer magnesium

The innermost electron shell of an atom can hold up to ______ electrons. - answer2

Which of these relationships is true of an uncharged atom? - answer The number of protons is equal to the number of electrons

What determines the types of chemical reactions that an atom participates in? answerThe number of electrons in the outermost electron shell

What is the atomic number of an atom that has 6 protons, 6 neutrons, and 6 electrons? - answer6

Which of these refers to atoms with the same atomic number but different atomic masses? - answerThese atoms are isotopes

Fluorine's atomic number is 9 and its atomic mass is 19. How many neutrons does fluorine have? - answer10

An uncharged atom of boron has an atomic number of 5 and an atomic mass of 11. How many protons does boron have? - answer5

What is the mass number of an ion with 107 electrons, 157 neutrons, and a +1 charge? - answer265

Which of the following subatomic particles always has a positive charge? answerproton

Changing the number of _____ would change an atom into an atom of a different element - answerprotons in an atom

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Wallace, Tietjen

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