

## Gizmos: Types of Reactions

### Activity B: Classifying reactions

Get the Gizmo ready:

- Turn off **Show summary** and **Show histograms**.



**Introduction:** Chemical equations show how **compounds** and **elements** react with one another. An element is a substance consisting of one kind of atom, such as aluminum (Al) or oxygen gas (O<sub>2</sub>). A compound is a substance made of more than one kind of atom, such as water (H<sub>2</sub>O) or table salt (NaCl).

**Question:** How are chemical reactions classified?

1. **Match:** Most chemical reactions can be classified as one of four types. Using the chemical equations in the Gizmo as a guide, match the following definitions to the type of reaction.

\_\_\_\_ One reactant is broken down into two or more products.

A. **Addition**

\_\_\_\_ Two or more reactants combine to form one product.

B. **Decomposition**

\_\_\_\_ Two compounds react to form two different compounds.

C. **Single replacement**

\_\_\_\_ A compound reacts with an element to form a new compound and a different element.

D. **Double replacement**

2. **Practice:** Balance each of the chemical equations below. (Some equations may already be in balance.) In the space to the right, classify the reaction as an *addition*, *decomposition*, *single replacement*, or *double replacement* reaction.



# Answer Sheets For Chemistry Gizmos

**Arshad Iqbal**



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**Popular Photography**, 2006-04

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