

Unit 3 Review - Atomic Structure & Periodic Table

Part 1: Answer the following questions.

1. a. Which scientist created the first modern atomic theory? John Dalton
 b. What was his theory?
 1. All matter is made of extremely small particles called atoms.
 2. All atoms of a given element are identical and atoms of a specific element are different from any other element
 3. Atoms cannot be created, divided into smaller particles or destroyed
 4. Different atoms combine in simple whole number ratios to form compounds
 5. In a chemical reaction, atoms are separated, combined or rearranged.
- c. What was his model of the atom called? Billiard Ball
- d. What was proven wrong and why?
 1. All atoms of a given element are identical – wrong b/c of the existence of isotopes
 2. Atoms cannot be divided into smaller particles – wrong b/c of the existence of electrons, protons and neutrons
2. a. Which scientist did the gold foil experiment? Rutherford
 b. What did he conclude from the gold foil experiment?
 1. Atoms are made up of mostly empty space
 2. Atoms have a small positive center call the nucleus
3. a. Who discovered the electron? J. J. Thomson
 b. What was the name of his experiment? Cathode Ray Tube Experiment
 c. What was his model of the atom called? Plum Pudding Model
4. What did Mendeleev do?
Arranged the periodic table in order of increasing atomic mass
5. a. Who arranged the **modern** periodic table? Moseley
 b. How is the modern periodic table arranged? Arranged in order of Atomic Number
6. a. What are the three subatomic particles? Give the symbols, charge, mass and location of each.

Particles	Symbol	Charge	Mass	Location
Proton	p^+	Positive	1 amu	Nucleus
Neutron	n^0	Neutral	1 amu	Nucleus
Electron	e^-	Negative	1/2000 amu	Electron Cloud

7. What is special about the proton? The proton IDENTIFIES THE ELEMENT
8. Fill in the blanks in the following:
 - a. Mass # = $p^+ + n^0$
 - b. Atomic # = # of protons
9. What is the difference between Average Atomic mass and mass #? (Average) Atomic Mass is found on the periodic table and is the weighted average of the isotopes. The Mass Number is the mass of one element and must be given or calculated by adding the protons and neutron of that element.
10. Which of the following are isotopes of the same element? How do you know? A, and D. They have the same Atomic Number (# of Protons)
 - a. $^{12}_5X$
 - b. $^{15}_5X$
 - c. $^{12}_6X$
 - d. $^{15}_5X$
11. Calculate the average atomic mass for Boron if it has two naturally occurring isotopes: boron -10 (abundance of 19.8%) and boron - 11 (abundance of 80.2%).

$$\left(10 \times \frac{19.8}{100} \right) + \left(11 \times \frac{80.2}{100} \right) = 10.802 \text{ amu}$$
12. Identify an element "X" by calculating the average atomic mass. X = 6 (abundance of 7.5%) and X= 7 (abundance 92.5%).

$$\left(6 \times \frac{7.5}{100} \right) + \left(7 \times \frac{92.5}{100} \right) = 6.925 \text{ amu} \quad \text{Which is Lithium}$$

Answer Key Unit 3 Review Electrons

Jeffrey C. Pommerville

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the author's name.

Answer Key Unit 3 Review Electrons:

Mosby's Radiation Therapy Study Guide and Exam Review Leia Levy, 2025-08-21 Reinforce your understanding of radiation therapy and prepare for the Registry exam Mosby's Radiation Therapy Study Guide and Exam Review Second Edition is both a study companion for Washington and Leaver's Principles and Practice of Radiation Therapy and a superior review for the ARRT Radiation Therapy Certification Exam This completely updated edition reflects the latest exam specifications and features an easy to read format that presents information in concise bullets and tables More than 2 000 total multiple choice questions in Registry format provide a realistic testing experience to prepare you for the real exam NEW Quality control procedures and guidelines for linear accelerators NEW Fractionation and protraction considerations UPDATED Content reflects the latest ARRT Radiation Therapy Certification Exam Specifications UPDATED Radiation Protection and Safety and Overview of Cancer Imaging and Management Modalities chapters offer the most current information in these key areas EXPANDED Charge capture and record keeping content enhances the Oncology Patient Care chapter EXPANDED Additional cross sectional images provide anatomy review and reinforce treatment planning concepts More than 2 000 multiple choice questions in Registry format in the text and on the companion Evolve website provide a realistic exam experience Complete coverage helps you prepare for the ARRT Radiation Therapy Certification Exam Content review in outline and tabular format provides a concise recap of the material you need to know to succeed on the exam Exercises at the end of each section offer engaging active review opportunities

Jacaranda Chemistry 2 VCE Units 3 and 4, 3e learnON and Print Robert Stokes, Angela Stubbs, Neale Taylor, Jason Bourke, Ben Williams, Maida Derbogosian, 2023-12-26 Developed by expert Victorian teachers for VCE students The NEW Jacaranda Chemistry VCE series continues to deliver curriculum aligned material that caters to students of all abilities Our expert author team of practising teachers and assessors ensures 100% coverage of the new VCE Chemistry Study Design 2023 2027 **Mosby's**

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Function of the Body 17th Edition introduces the typical structure and function of the human body and describes what the body does to maintain homeostasis The book shows how structure fits function using clinical examples to reinforce A P concepts and featuring hundreds of photos and micrographs for realistic visual detail Written by a team of experts led by Kevin Patton this text includes an Evolve website packed with animations audio pronunciations review questions and other interactive learning resources NEW Updated content is added and new line art and photos ensure wider representation of skin color sex age body type and cultural diversity NEW Inclusive terminology reduces the emphasis on eponyms for example the term normal is more carefully used to avoid implying that healthy conditions outside the average are abnormal NEW The latest scientific thinking introduces or expands upon emerging core concepts such as the human microbiome with a new diagram illustrating the changes in the microbiome throughout the human life cycle Clear conversational writing style is paired with chunked content which breaks down the material into smaller bite sized bits of information that are easier to read and understand More than 400 full color photos micrographs and drawings illustrate the diversity and detail of the human body Language of Science and Medicine lists in each chapter includes key terms pronunciations and word parts to highlight new or complex medical terminology NEW Updated Connect It boxes refer you to articles on Evolve that integrate concepts and discuss the latest clinical developments and scientific research showing the big picture of human structure and function NEW Updated Science Application boxes discuss possible career paths within the context of a diversity of historical figures and their life stories NEW Quick Guide to the Language of Science and Medicine is added to Evolve helping you learn medical terminology without the need for a separate textbook UNIQUE 22 page Clear View of the Human Body insert allows you to peel back the layers of the human body both male and female by flipping through full color semi transparent pages Student friendly features make learning easier with chapter outlines chapter objectives key terms study hints frequent Quick Check questions chapter summaries review questions critical thinking questions chapter tests and more Boxed sidebars include Health and Well Being Clinical Application Research Issues and Trends and Science Applications to help you apply concepts and develop critical thinking skills Resources on the Evolve website include animations audio summaries audio pronunciations the Body Spectrum anatomy coloring book review questions and FAQs with answers from the authors

FCS Electronics L2 ,2007 **Chemistry**³ Andrew Burrows,Andrew Parsons,Gwen Pilling,Gareth Price,2013-03-21 New to this Edition **Concepts in Biology' 2007 Ed.2007 Edition** , Resources in Education ,1991-04 CUET (UG) 20 Combined Mock Test Papers Science (PCB) Stream (English, Physics, Chemistry, Biology, General Test) For 2025 Exam Oswaal Editorial Board,2024-09-05 The Central Universities Entrance Test CUET UG has emerged as a crucial examination for students aspiring to secure admission into prestigious universities across India With the competition intensifying each year thorough preparation and strategic practice have become essential for success In 2024 The Common University Entrance Test CUET had conducted the assessment test for admission to UG programme across 44 Central Universities at

354 centres in India and at 25 centres abroad In 2024 nearly 15 lakh candidates registered for the CUET UG Exam Oswaal Combined Mock Test Papers have been meticulously designed to simulate the syllabus and actual exam pattern issued by the NTA ensuring that students can practice in an environment similar to the one they will face on the day of the examination Each test paper covers a wide range of topics carefully aligned with the CUET syllabus helping students assess their knowledge hone their problem solving skills and improve their time management in the CBT test A Hybrid mode Pen Paper CBT question paper is divided into four sections Sections IA IB contain language specific questions Section II comprises domain related questions and Section III contains questions on general topics All questions are MCQ based This book of Combined Mock Test Papers for CUET UG aims to provide a comprehensive and well structured resource for students preparing in various streams including Commerce English Accountancy Business Studies Economics and General Test Humanities English History Geography Political Science and General Test Science PCB PCM English Physics Chemistry Biology Mathematics and General Test Here's how this book benefits you 100% updated with the Latest Solved Paper 2024 for unmatched accuracy relevance 20 Mock test papers 4 sets in each subject for extensive practice 1200 Questions approx for comprehensive coverage Answer key with explanations to ensure in depth concept clarity OMR Sheet simulates the real exam experience We believe that consistent practice and familiarity with the exam format are key to excelling in CUET UG By working through these Mock Test Papers students will not only build a strong foundation in their respective subjects but also develop the confidence needed to tackle the exam with ease We hope this book serves as a valuable companion in your preparation journey We wish all aspiring students the very best of luck in their endeavours and future academic pursuits

Journal of Research of the National Bureau of Standards United States. National Bureau of Standards, 1988

Electricity and Magnetism, 2005 **Atoms, Molecules & Elements: The Periodic Table Gr. 5-8** George

Graybill, 2015-10-01 This is the chapter slice The Periodic Table from the full lesson plan Atoms Molecules Elements Young scientists will be thrilled to explore the invisible world of atoms molecules and elements Our resource provides ready to use information and activities for remedial students using simplified language and vocabulary Students will label each part of the atom learn what compounds are and explore the patterns in the periodic table of elements to find calcium Ca chlorine Cl and helium He through hands on activities These and more science concepts are presented in a way that makes them more accessible to students and easier to understand Written to grade and using simplified language and vocabulary and comprised of reading passages student activities crossword word search comprehension quiz and color mini posters our resource can be used effectively for test prep and your whole class All of our content is aligned to your State Standards and are written to Bloom's Taxonomy and STEM initiatives *Journal of Research of the National Bureau of Standards*, 1984

Atoms, Molecules & Elements: What Are Elements? Gr. 5-8 George Graybill, 2015-10-01 This is the chapter slice What Are Elements from the full lesson plan Atoms Molecules Elements Young scientists will be thrilled to explore the

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 - Chapter 4: Answer Key Unit 3 Review Electrons in Specific Contexts
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