

SCHOOLNAME:-

STUDENT'S NAME:-

CLASS:-

SENIOR CLASS TWO (S.S.S. 2)

SUBJECT:-

PHYSICS 1ST - 3RD TERM EXAM

18.
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SECTION A: FIRST TERM QUESTIONS

1. A phenomenon which is used to describe the process by which a substance changes from solid to gas is called (a) evaporation (b) freezing (c) sublimation (d) melting (e) vaporization
2. A phenomenon which is used to describe the process by which a substance changes from liquid to vapour is called (a) evaporation (b) freezing (c) sublimation (d) melting (e) vaporization
3. The constant temperature at which a substance changes state from liquid to solid is called (a) freezing point (b) melting point (c) boiling point (d) heating point
4. Melting point is a temperature at which a solid substance has its bond broken such that it now flows as liquid. (a) freezing point (b) melting point (c) boiling point (d) heating point
5. What effect will reducing the surface pressure of a liquid have on its boiling point? (a) increase its boiling point (b) reduces its boiling point (c) it does not have effect (d) increases its boiling point by 5°C (e) none of the above
6. Which of the following statement is true about evaporation (a) evaporation occurs at all temperature except at boiling points (b) evaporation only occurs at boiling point (c) evaporation occurs at all temperature (d) all of the above (e) none of the above
7. Impurities change the boiling points of liquid. So, salt added to water will (a) increase its boiling point (b) reduces its boiling point (c) it does not have effect (d) reduces its boiling point by 5°C (e) none of the above
8. Which of the following is the correct SI unit of specific latent heat? (a) J/kg (b) J/kg·K (c) J/kg·K (d) J/kg (e) J/kg
9. The heat required to change 1kg of ice at 0°C to water at the same temperature is called (a) specific latent heat of vaporization (b) specific heat capacity (c) specific latent heat of fusion (d) heat capacity (e) specific heat capacity
10. The equation $PV^{\gamma}T^{\gamma} = \text{constant}$ is Boyle's law if (a) $x=0, y=0, z=1$ (b) $x=1, y=0, z=0$ (c) $x=1, y=1, z=0$ (d) $x=1, y=1, z=1$ (e) $x=1, y=1, z=-1$
11. A fixed mass of gas of volume 600cm³ at a temperature of 27°C is cooled at constant pressure to a temperature of 0°C. What is the change in volume? (a) 54cm³ (b) 273cm³ (c) 300cm³ (d) 546cm³ (e) 600cm³
12. A fixed mass of gas occupies a volume of 20cm³ at a fixed pressure of 700mmHg. Assuming that the temperature remains constant, what will be the volume of the gas at 750mmHg? (a) 2.5cm³ (b) 15.5cm³ (c) 18.7cm³ (d) 21.4cm³ (e) 72.5cm³
13. _____ is an agent that changes or tends to change the state of rest or of uniform motion on a straight line of a body (a) force (b) motion (c) friction (d) movement
14. _____ is the movement of an object from one place to another at a particular time (a) force (b) motion (c) friction (d) movement
15. Force that are in contact with the body they affect are called _____ (a) contact force (b) force field (c) non-contact force (d) gravitational force
16. Force whose sources does not required contact with the body on which they act is called _____ (a) contact force (b) force field (c) non-contact force (d) gravitational force
17. If 1.13×10^4 J of heat energy is required to convert 15kg of steam to water. Calculate the specific latent heat of vaporization of steam (a) 7533J/kg (b) 7500J/kg (c) 7870J/kg (d) 8245J/kg

18. The law that states that the pressure of a fixed mass of gas varies inversely as the volume at constant temperature is called _____ (a) Boyle's law (b) Charles's law (c) pressure law (d) general gas law
19. _____ states that for a fixed mass of gas at constant pressure, the volume is proportional to its absolute temperature (a) Boyle's law (b) Charles's law (c) pressure law (d) general gas law
20. When two bodies collide and then move with different velocities (a) elastic collision (b) inelastic collision (c) momentum (d) displacement
21. Which of the following is not an effect of heat? A. expansion B. contraction C. change of state. D. increase in weight
22. Temperature can be measured in the following units except A. degree Celsius B. kelvin C. degree Fahrenheit D. centigrade-meter
23. When heat is extracted from water it changes from liquid to gas. What is this process called _____ (a) vaporization (b) evaporation (c) condensation
24. The process of by which a gas is converted to a solid is called _____ (a) sublimation (b) deposition (c) vaporize
25. Fahrenheit is a unit for _____ (a) heat (b) gas (c) temperature (a) water
26. _____ of a substance is defined as the amount of heat required to raise the temperature of 1 gram of a substance through 1°C. (a) special heat capacity (b) specific heat capacity (c) molar heat capacity
27. _____ of a substance is defined as the amount of heat required to raise the temperature of 1 mole of a substance through 1° C (a) special heat capacity (b) specific heat capacity (c) molar heat capacity
28. There are _____ types of heat capacities (a) 2 (b) 4 (c) 3
29. _____ The heat supplied to a system to raise its temperature through 1° C keeping the volume of the system constant is called (a) heat capacity at constant pressure (b) heat capacity at constant volume (c) thermodynamics
30. The heat supplied to a system to raise its temperature through 1° C keeping the external pressure constant is called _____ (a) heat capacity at constant pressure (b) heat capacity at constant volume (c) thermodynamics
31. An electric heater rated 12V is used to heat 450g of water when a current of 5A was passed through it. What is the final temperature after 30minutes? [Specific capacity of water = 4200 J/kg·K] (a) 57k (b) 20k (c) 57k (d) 80k (e) 40k
32. If the pressure of the vapor on top of an enclosed liquid is equal to the atmospheric pressure, what will be the temperature of the liquid enclosed? (a) room temperature (b) boiling point (c) freezing point (d) standard temperature
33. 200g of water at 90°C is mixed with 100g of water at 30°C. What is the final temperature? (a) 50°C (b) 60°C (c) 70°C (d) 80°C
34. Which of the following is not used to determine the heat content of a body? (a) mass of the body (b) volume of the body (c) specific heat capacity of the body (d) temperature of the body (e) all of the above
35. A _____ is an instrument used in measuring the quantity of heat (a) Barometer (b) Aerometer (c) anemometer (d) calorimeter
36. _____ is defined as the process of reducing heat hot to the surrounding (a) lagging (b) calorimeter (c) regnant (d) capacity
37. It is the spectrum of sunlight in nature. It is formed due to the dispersion of sunlight

2015 Ss2 Mock Exam

Ying-Ying Zheng



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The Top Books of the Year 2015 Ss2 Mock Exam The year 2023 has witnessed a remarkable surge in literary brilliance, with numerous compelling novels captivating the hearts of readers worldwide. Lets delve into the realm of top-selling books, exploring the engaging narratives that have charmed audiences this year. The Must-Read : Colleen Hoover "It Ends with Us" This poignant tale of love, loss, and resilience has gripped readers with its raw and emotional exploration of domestic abuse. Hoover masterfully weaves a story of hope and healing, reminding us that even in the darkest of times, the human spirit can triumph. Uncover the Best : Taylor Jenkins Reids "The Seven Husbands of Evelyn Hugo" This spellbinding historical fiction novel unravels the life of Evelyn Hugo, a Hollywood icon who defies expectations and societal norms to pursue her dreams. Reids absorbing storytelling and compelling characters transport readers to a bygone era, immersing them in a world of glamour, ambition, and self-discovery. Discover the Magic : Delia Owens "Where the Crawdads Sing" This evocative coming-of-age story follows Kya Clark, a young woman who grows up alone in the marshes of North Carolina. Owens crafts a tale of resilience, survival, and the transformative power of nature, entrancing readers with its evocative prose and mesmerizing setting. These popular novels represent just a fraction of the literary treasures that have emerged in 2023. Whether you seek tales of romance, adventure, or personal growth, the world of literature offers an abundance of engaging stories waiting to be discovered. The novel begins with Richard Papen, a bright but troubled young man, arriving at Hampden College. Richard is immediately drawn to the group of students who call themselves the Classics Club. The club is led by Henry Winter, a brilliant and charismatic young man. Henry is obsessed with Greek mythology and philosophy, and he quickly draws Richard into his world. The other members of the Classics Club are equally as fascinating. Bunny Corcoran is a wealthy and spoiled young man who is always looking for a good time. Charles Tavis is a quiet and reserved young man who is deeply in love with Henry. Camilla Macaulay is a beautiful and intelligent young woman who is drawn to the power and danger of the Classics Club. The students are all deeply in love with Morrow, and they are willing to do anything to please him. Morrow is a complex and mysterious figure, and he seems to be manipulating the students for his own purposes. As the students become more involved with Morrow, they begin to commit increasingly dangerous acts. The Secret History is a brilliant and gripping novel that will keep you speculating until the very end. The novel is a cautionary tale about the dangers of obsession and the power of evil.

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