

The Periodic Table

H																	He
Li	Be											B	C	N	O	F	Ne
Na	Mg											Al	Si	P	S	Cl	Ar
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr
Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe
Cs	Ba	La	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn
Fr	Ra	Ac	Rf	Ha	Sg	Uns	Uno	Une									
		Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu		
		Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr		

- The elements in periods 2 through 7 that are to the left of the zigzagged line are metals. List five metals.

- The elements in periods 1 through 6 that are to the right of the zigzagged line are nonmetals. List five nonmetals.

- The elements that border either side of the zigzagged line are metalloids. Metalloids have properties of both metals and nonmetals. Name two metalloids.

- The alkali metals are in group 1. They have only one electron in their outer shell. They are very reactive and have a low melting point. Shade the alkali metals blue.
- The noble gases are found in group 18. Their outer energy level is filled, and they are very nonreactive, colorless gases. Shade the noble gases red.
- Halogens have seven electrons in the outer shell. They are located in group 17 on the periodic table. Halogens react with metals to form salts. Shade the halogens green.
- The transition elements are found in groups 3 through 12 and periods 4 through 7. These elements have either one or two electrons in the outer energy level. They are often used to form alloys because they are hard and have high melting points. Shade the transition elements orange.
- Alkaline earth metals are located in group 2. They have two electrons in their outer energy level. Shade the alkaline metals purple.

Answer Sheet For The Periodic Table Most Wanted

Marc Rochkind



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Girolami, 2021-07-04 This book provides an overview of the origins and evolution of the periodic system from its prehistory to the latest synthetic elements and possible future additions The periodic system of the elements first emerged as a comprehensive classificatory and predictive tool for chemistry during the 1860s Its subsequent embodiment in various versions has made it one of the most recognizable icons of science Based primarily on a symposium titled 150 Years of the Periodic Table and held at the August 2019 national meeting of the American Chemical Society this book describes the origins of the periodic law developments that led to its acceptance chemical families that the system struggled to accommodate extension of the periodic system to include synthetic elements and various cultural aspects of the system that were celebrated during the International Year of the Periodic Table **The Publishers' Circular and Booksellers'**

Record of British and Foreign Literature, 1902 Ebook: Chemistry: The Molecular Nature of Matter and Change

Silberberg, 2015-01-16 Ebook Chemistry The Molecular Nature of Matter and Change Mission to Algiers Cameron R.

Hume, 2006-01-01 Mission to Algiers relates how U S policy grappled with an Islamist insurgency and promoted change and stability in a key Arab country It describes the course of events the challenges inherent in operating a U S mission at a time of crisis in a Muslim country and lessons learned **i write HE DICTATES - Revenge** Arnold Iwin Francis, 2024-02-09

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