GRADE 12 PRESCRIBED EXPERIMENT 1: PREPARATION OF ESTERS
NAME OF LEARNER
DATE
SCHOOL
TOTAL MARKS::50

1. Title

Preparation and identification of Esters

2. Aim of the experiment

* To synthesize/prepare esters, methylbutanoate and ethylbutanoate.

- To write the chemical equations for the formation of methylbutanoate and ethylbutanoate using structural formulas.
- 3. Materials/Apparatus

Equipment/ Glasswares
2 test tubes
Water soluble marker
Test tube rack
Eye dropper
2 x 250 ml beakers
Hot plate
10 ml graduated cylinder
Thermometer
Lab apron & Safety goggles
Water bath

Chemical / Reagents Methanol Ethanol Concentrated sulphuric acid Butanoic acid

4. Safety Precautions

- Concentrated sulphuric acid is a strong exidizing agent and highly corrosive. It is
 dangerous and can burn skin, eyes, and clothing very badly. As oid contact with skin
 or clothes. Wear gloves and proper eye protection when using this substance. Use only
 in a fume hood. Use exactly as directed. If it is spilled, wash immediately before the
 acid has a chance to cause a burn, and inform the instructor.
- Butanoic acid (all organic liquid acids) is toxic and corrosive to skin, eyes, and clothing. While working on this experiment, wear safety goggles, full face shield, gloves, and lab apron. Wash spills and splashes off your skin and clothing immediately, using plenty of water.
- All burners and other flames in the laboratory must be extinguished before you start this experiment.
 - Methanol is highly flammable. It is poisonous if swallowed. Its vapour is harmful
 to the eyes, lungs and skin and other organs.
 - Ethanol is flammable, poisonous and its toxic.
- Detect odours of these esters with caution because vapors produced may be harmful. Breathing the vapours of some of these esters can cause sore throat, dizziness, headache, and drowsiness. Hold the test tube 30 cm away and 15 cm below your nose. Waft a small amount of vapor or odour from the ester toward your nose, sniffing cautiously, once or twice. Do not breathe deeply while sniffing.

Christian G. Meyer

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