



Name: _____ Date: _____

Student Exploration: Nuclear Decay

Background: Alpha particles, beta particles, gamma rays, and neutrons are particles that are emitted from unstable nuclei. Alpha particles consist of two protons and two neutrons, beta particles consist of one electron or one positron, gamma rays consist of high-energy photons, and neutrons consist of one neutron.

Prior Knowledge Questions (What do you already know about this topic?)

1. What are some properties of atoms, isotopes, and radioactive materials? (What do you know about them?)

Particle	Composition	Charge	Approximate mass
Alpha	2 protons, 2 neutrons	+2	4 u
Beta	1 electron or 1 positron	-1 or +1	1/1836 u
Gamma	High-energy photon	0	0

1. The **atomic number** of an atom is equal to the total number of protons in the nucleus.

2. Alpha particles have 2 protons and 2 neutrons. What is the mass number of the parent? (What is the mass number of the daughter?)

3. The **mass number** of an atom is equal to the total number of protons and neutrons in the nucleus.

4. Alpha particles have 2 protons and 2 neutrons. What is the mass number of the daughter?

Alpha Particles

Alpha particles are made of two protons and two neutrons. They are emitted from the nucleus of a radioactive atom.

Background: The decay of a radioactive material is a random process. It is not possible to predict when a particular atom will decay.

Alpha particles are emitted from the nucleus of a radioactive atom. They are made of two protons and two neutrons. They have a mass of 4 atomic mass units and a charge of +2.



1. Alpha particles have 2 protons and 2 neutrons. What is the mass number of the parent? (What is the mass number of the daughter?)

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4. Alpha particles have 2 protons and 2 neutrons. What is the mass number of the daughter?

Background: Alpha particles are made of two protons and two neutrons. They have a mass of 4 atomic mass units and a charge of +2.



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