

Assignment #2 – Compound Names and Formulas
Multi-valent ions only

A. Write the correct formula for the following compounds, all of which have been named using the modern Roman Numeral Method. The ionic charge is given after the first element (metallic) in Roman Numerals.

1. copper (II) oxide CuO
2. mercury (I) oxide Hg₂O
3. gold (III) chloride AuCl₃
4. thallium (III) bromide TlBr₃
5. bismuth (V) oxide Bi₂O₅
6. terbium (III) oxide Tb₂O₃
7. uranium (VI) oxide UO₃
8. protactinium (IV) bromide PaBr₄
9. cerium (III) oxide Ce₂O₃
10. arsenic (V) sulphide As₂S₅

11. manganese (III) oxide Mn₂O₃
12. vanadium (II) bromide VBr₂
13. niobium(V) oxide Nb₂O₅
14. titanium (III) oxide Ti₂O₃
15. titanium (III) nitride TiN
16. iron (II) oxide FeO
17. cobalt (II) phosphide Co₃P₂
18. tin (II) oxide SnO
19. thulium (II) bromide TmBr₂
20. copper (II) bromide CuBr₂

B. Name the compound using the Roman Numeral Method and balance the ionic charges.

1. SnCl₄ Tin (IV) Chloride
2. BiBr₅ Bismuth (V) Bromide
3. PoO₂ Polonium (IV) Oxide
4. PbI₂ Lead (II) Iodide
5. HgO Mercury (II) Oxide
6. HgCl Mercury (I) Chloride
7. Au₂O₃ Gold (III) Oxide
8. FeCl₂ Iron (II) Chloride

9. PdF₄ Palladium (IV) Fluoride
10. Os₂O₃ Osmium (III) Oxide
11. MoBr₂ Molybdenum (II) Bromide
12. VCl₅ Vanadium (V) Chloride
13. Mn₂O₃ Manganese (III) Oxide
14. CoO Cobalt (II) Oxide
15. Np₂O₃ Neptunium (III) Oxide
16. V₂O₅ Vanadium (V) Oxide

Answers For Writing Formulas From Names

Martha Mackin



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Table of Contents Answers For Writing Formulas From Names

1. Understanding the eBook Answers For Writing Formulas From Names
 - The Rise of Digital Reading Answers For Writing Formulas From Names
 - Advantages of eBooks Over Traditional Books
2. Identifying Answers For Writing Formulas From Names
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answers For Writing Formulas From Names
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answers For Writing Formulas From Names
 - Personalized Recommendations
 - Answers For Writing Formulas From Names User Reviews and Ratings

- Answers For Writing Formulas From Names and Bestseller Lists
- 5. Accessing Answers For Writing Formulas From Names Free and Paid eBooks
 - Answers For Writing Formulas From Names Public Domain eBooks
 - Answers For Writing Formulas From Names eBook Subscription Services
 - Answers For Writing Formulas From Names Budget-Friendly Options
- 6. Navigating Answers For Writing Formulas From Names eBook Formats
 - ePub, PDF, MOBI, and More
 - Answers For Writing Formulas From Names Compatibility with Devices
 - Answers For Writing Formulas From Names Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answers For Writing Formulas From Names
 - Highlighting and Note-Taking Answers For Writing Formulas From Names
 - Interactive Elements Answers For Writing Formulas From Names
- 8. Staying Engaged with Answers For Writing Formulas From Names
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answers For Writing Formulas From Names
- 9. Balancing eBooks and Physical Books Answers For Writing Formulas From Names
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answers For Writing Formulas From Names
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Answers For Writing Formulas From Names
 - Setting Reading Goals Answers For Writing Formulas From Names
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Answers For Writing Formulas From Names
 - Fact-Checking eBook Content of Answers For Writing Formulas From Names
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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