

Buffers Keep the Balance

Data Table 1. Model Carbonate Blood Buffer

Table 1	Test Tube Number					
	1	2	3	4	5	6
Solution	Carbonic Acid (Reference)	Model Carbonate Blood Buffer	Water (Control)	Model Carbonate Blood Buffer	Water (Control)	Sodium Bicarbonate (Reference)
Bromothymol blue indicator color	yellow 6.0	greenish 7.2 blue	green 6.0	greenish 7.0 blue	green 6.0	blue
Initial pH value	6.0	7.2	6.0	7.0	6.0	8.0
Number of drops of HCl required to convert solutions 2 and 3 to yellow (acid reference color)	NA	16	5	NA	NA	NA
Number of drops of NaOH required to convert solutions 4 and 5 to blue (base reference color)	NA	NA	NA	2	2	NA

Data Table 2. Effect of HCl on Biological Phosphate Buffers

	Estimated pH (pH paper)	Universal Indicator Color			
		Initial	After 1 drop of HCl	After 5 drops of HCl	After 10 drops of HCl
Water (control)	6.0	6.0	6.0	NA	NA
Phosphate Buffer A	6.0	6.0	6.0	6.0	6.0
Phosphate Buffer B	6.4	6.4	6.4	6.0	6.0

Data Table 3. Effect of NaOH on Biological Phosphate Buffers

	Estimated pH (pH paper)	Universal Indicator Color			
		Initial	After 1 drop of NaOH	After 5 drops of NaOH	After 10 drops of HCl
Water (control)	6.0	6.0	7.0	NA	NA
Phosphate Buffer A	6.0	6.0	6.8	7.0	8.0
Phosphate Buffer B	6.4	6.4	7.0	7.0	7.0

Answers Buffers Keep The Balance Lab Answers

SJ Ball



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Biochemistry in the Lab Benjamin F. Lasseter, 2019-09-30 Most lab manuals assume a high level of knowledge among biochemistry students as well as a large amount of experience combining knowledge from separate scientific disciplines Biochemistry in the Lab A Manual for Undergraduates expects little more than basic chemistry It explains procedures clearly as well as giving a clear explanation of the theoretical reason for those steps Key Features Presents a comprehensive approach to modern biochemistry laboratory teaching together with a complete experimental experience Includes chemical biology as its foundation teaching readers experimental methods specific to the field Provides instructor experiments that are easy to prepare and execute at comparatively low cost Supersedes existing older texts with information that is adjusted to modern experimental biochemistry Is written by an expert in the field This textbook presents a foundational approach to modern biochemistry laboratory teaching together with a complete experimental experience from protein purification and characterization to advanced analytical techniques It has modules to help instructors present the techniques used in a time critical manner as well as several modules to study protein chemistry including gel techniques enzymology crystal growth unfolding studies and fluorescence It proceeds from the simplest and most important techniques to the most difficult and specialized ones It offers instructors experiments that are easy to prepare and execute at comparatively low cost

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