

| Question Number | Acceptable Answers                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                         | Reject  | Mark      |       |     |       |   |       |      |        |       |   |      |     |        |       |   |       |      |          |       |   |       |      |           |       |  |   |
|-----------------|------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|---------|-----------|-------|-----|-------|---|-------|------|--------|-------|---|------|-----|--------|-------|---|-------|------|----------|-------|---|-------|------|-----------|-------|--|---|
| 23(a)           | <table><tr><th>Element</th><th>%</th><th></th><th>mol</th><th>Ratio</th></tr><tr><td>C</td><td>40.44</td><td>÷ 12</td><td>= 3.37</td><td>2.999</td></tr><tr><td>H</td><td>7.87</td><td>÷ 1</td><td>= 7.87</td><td>7.004</td></tr><tr><td>O</td><td>35.96</td><td>÷ 16</td><td>= 2.2475</td><td>2.000</td></tr><tr><td>N</td><td>15.73</td><td>÷ 14</td><td>= 1.12357</td><td>1.000</td></tr></table> <p style="text-align: center;">(1)                      (1)</p> <p>Empirical formula = C<sub>3</sub>H<sub>7</sub>O<sub>2</sub>N (1)<br/>Symbols in any order<br/>Stand alone mark<br/>No TE on incorrect ratio<br/>I IGNORE significant figure and rounding errors except 1sf in mole calculation</p> | Element | %         |       | mol | Ratio | C | 40.44 | ÷ 12 | = 3.37 | 2.999 | H | 7.87 | ÷ 1 | = 7.87 | 7.004 | O | 35.96 | ÷ 16 | = 2.2475 | 2.000 | N | 15.73 | ÷ 14 | = 1.12357 | 1.000 |  | 3 |
| Element         | %                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                          |         | mol       | Ratio |     |       |   |       |      |        |       |   |      |     |        |       |   |       |      |          |       |   |       |      |           |       |  |   |
| C               | 40.44                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                      | ÷ 12    | = 3.37    | 2.999 |     |       |   |       |      |        |       |   |      |     |        |       |   |       |      |          |       |   |       |      |           |       |  |   |
| H               | 7.87                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                       | ÷ 1     | = 7.87    | 7.004 |     |       |   |       |      |        |       |   |      |     |        |       |   |       |      |          |       |   |       |      |           |       |  |   |
| O               | 35.96                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                      | ÷ 16    | = 2.2475  | 2.000 |     |       |   |       |      |        |       |   |      |     |        |       |   |       |      |          |       |   |       |      |           |       |  |   |
| N               | 15.73                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                      | ÷ 14    | = 1.12357 | 1.000 |     |       |   |       |      |        |       |   |      |     |        |       |   |       |      |          |       |   |       |      |           |       |  |   |

| Question Number | Acceptable Answers                                                                                                       | Reject | Mark     |
|-----------------|--------------------------------------------------------------------------------------------------------------------------|--------|----------|
| <b>23(b)(i)</b> | Peak at $m/e = 89$ labelled $\text{M}^+$<br><br>ALLOW<br>Any clear label e.g. $\text{C}_3\text{H}_7\text{O}_2\text{N}^+$ |        | <b>1</b> |

| Question Number  | Acceptable Answers                                                                                                                                                                                | Reject                                                                                 | Mark     |
|------------------|---------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|----------------------------------------------------------------------------------------|----------|
| <b>23(b)(ii)</b> | $M_r = 89 = M_r(\text{C}_3\text{H}_7\text{O}_2\text{N})$ so molecular formula is $\text{C}_3\text{H}_7\text{O}_2\text{N}$<br>Symbols in any order<br><br>IGNORE structural and displayed formulae | Answer with no explanation<br>$M_r = 90$<br>$\text{C}_3\text{H}_7\text{O}_2\text{N}^+$ | <b>1</b> |

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**Terry C. Jones**



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