

66. which radiation are of very high frequency
a. I.R. b. X-ray c. Visible rays d. U.V
67. Poise is equal to.....
a. $1000\text{gcm}^{-1}\text{sec}^{-1}$ b. $100\text{Kgm}^{-1}\text{sec}^{-1}$ c. $10\text{Kgm}^{-1}\text{sec}^{-1}$ d. $0.1\text{Kgm}^{-1}\text{sec}^{-1}$
68. Isomorphous substance have same
a. B.P b. M.P c. atomic point d. lattice point
69. In a process 200J of heat is absorbed by a system while 300J work done by system. The change in internal energy
a. 200J b. 50J c. 100J d. 300J
70. Molecular orbital of lower energy than atomic orbital is called.....
a. Bonding orbital b. antibonding orbital c. Atomic orbital d. Non-bonding orbital
71. A sample of gas occupies 50dm³ at S.T.P assuming constant pressure what volume will the gas occupy if temp is double.....
a. 25dm³ b. 50dm³ c. 200dm³ d. 2500dm³
72. Example of trigonal system is... a. BaSO₄ b. AgNO₃ c. ZnSO₄ d. SiO₂ e. SnO₂
73. Four container of equal volume are filled as follows I. 2gm of H₂ at 0°C II. 1g of H₂ at 27°C
III. 24gm of O₂ at 0°C IV. 36gm O₂. Which container is at greatest pressure.?
a. I b. II c. III d. IV
74. All of the following are non polar bond except. a. O₂ b. O₃ c. N₂ d. HF e. H₂
75. water has maximum density at temperature of... ? a. -4°C b. 4°C c. -20°C d. 100°C e. 10°C
76. Phenol phthalein is an indicator used for a. strong acid & weak base b. strong base & weak acid
c. Strong acid & strong base d. weak acid & weak base.
77. which of the following will be the most electrically conductive....
a. sugar dissolved in water b. salt water c. salt dissolved in an organic compound
d. an oil & water mixture
78. which of the following is an element
a. tap water b. graphite c. sea water d. Brass e. Sulphuric acid.
79. To what volume must 10.0ml of 5.00M HCl be diluted to make a 0.5M HCl solution.....
a. 1 ml b. 50ml c. 100ml d. 500ml e. 5000ml
80. The property being able to add enthalpies is based on the
a. law of conservation of heat b. first law of thermodynamics c. law of constant composition d. E=mc²
e. Hess's Law
81. If ΔH of a reaction is - 500Kcal/mol it indicates the reaction...
a. Endothermic b. exothermic c. unstable d. in need of catalyst e. N.O.T
82. Which of the following is most likely to increase the rate of a reaction
a. decreasing the temperature b. reducing the surface area of reactant
c. decreasing the concentration of reactant in the reaction vessel d. reducing activation energy
e. +ve catalyst.

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Table of Contents 1sec Sample Paper

1. Understanding the eBook 1sec Sample Paper
 - The Rise of Digital Reading 1sec Sample Paper
 - Advantages of eBooks Over Traditional Books
2. Identifying 1sec Sample Paper
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an 1sec Sample Paper
 - User-Friendly Interface
4. Exploring eBook Recommendations from 1sec Sample Paper
 - Personalized Recommendations
 - 1sec Sample Paper User Reviews and Ratings
 - 1sec Sample Paper and Bestseller Lists
5. Accessing 1sec Sample Paper Free and Paid eBooks

- 1sec Sample Paper Public Domain eBooks
 - 1sec Sample Paper eBook Subscription Services
 - 1sec Sample Paper Budget-Friendly Options
6. Navigating 1sec Sample Paper eBook Formats
- ePub, PDF, MOBI, and More
 - 1sec Sample Paper Compatibility with Devices
 - 1sec Sample Paper Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of 1sec Sample Paper
 - Highlighting and Note-Taking 1sec Sample Paper
 - Interactive Elements 1sec Sample Paper
8. Staying Engaged with 1sec Sample Paper
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers 1sec Sample Paper
9. Balancing eBooks and Physical Books 1sec Sample Paper
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection 1sec Sample Paper
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine 1sec Sample Paper
- Setting Reading Goals 1sec Sample Paper
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of 1sec Sample Paper
- Fact-Checking eBook Content of 1sec Sample Paper
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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