

Chemistry, Part 1

Module 14 Notes



Big Question

How are metals used in chemical reactions? What products are produced?

Big Idea

Using chemical analysis, you can identify the products of a reaction between elements and other substances. A reaction between an element and a compound is used in chemistry as an analytical analysis with chemical reactions.

Learning Objectives

After completing this module, students will be able to:

- Use the law of conservation of mass to balance chemical reactions.
- Use chemical equations to describe processes of reactions.
- Calculate the mass of a substance.
- Calculate the mass of a substance.

Module 14 Objectives

Learn to use the law of conservation of mass to balance chemical reactions.

Fill in the blanks.

1. The law of conservation of mass states that the mass of the reactants is equal to the mass of the products.
2. The law of conservation of mass states that the mass of the reactants is equal to the mass of the products.
3. The law of conservation of mass states that the mass of the reactants is equal to the mass of the products.

Chemical Reactions and Equations

For the first reaction, the reactants are hydrogen and oxygen.

Hydrogen reacts with oxygen to form water.

Hydrogen reacts with oxygen to form water.

Hydrogen reacts with oxygen to form water.

Hydrogen reacts with oxygen to form water.

Module 14

Learn to use the law of conservation of mass to balance chemical reactions.

Fill in the blanks.

1. The law of conservation of mass states that the mass of the reactants is equal to the mass of the products.
2. The law of conservation of mass states that the mass of the reactants is equal to the mass of the products.

Answers For Chemistry Module 14

Mr. Rohit Manglik



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