

Experiment 2 - Gravimetric Analysis of a Chloride Salt (G)

Purpose:

Determining quantitative value of chloride in an unknown soluble salt by showing and explaining common methods in gravimetric analysis.

Theory:

- Reaction between silver ion and chloride ion to form silver chloride as an equation:



- AgCl solubility in water and the K_{sp} :

Silver chloride is very insoluble, so we can assume that the above reaction goes to completion.

$$K_{sp} = 1.6 \times 10^{-10}$$

- AgCl is precipitated with an acidic solution to reduce likelihood of acid ions interfering with results. These ions will form co-precipitates with silver in the acid, due to these co-precipitates, the results would be higher.
- To prevent the precipitate from going through the filtered paper, it is heated to coagulate it into a different texture that's easier to work with. Heating the precipitate makes it coagulate, therefore, it is harder for the precipitate to go through the filtered paper.
- Dry silver chloride decomposes into silver and chlorine in the presence of light:



- If photodecomposition takes place in air, results will be lower, since chlorine will be released in the form of gas. If photochemical decomposition occurs in the presence of excess silver ions in aqueous solution, it will result in higher results.

Analysis Of An Unknown Salt Lab Report

Jo Allan Beran



Analysis Of An Unknown Salt Lab Report:

Laboratory Manual for Principles of General Chemistry Jo Allan Beran, 2010-11-01 This new edition of the Beran lab manual emphasizes chemical principles as well as techniques The manual helps students understand the timing and situations for the various techniques The Beran lab manual has long been a market leading lab manual for general chemistry Each experiment is presented with concise objectives a comprehensive list of techniques and detailed lab intros and step by step procedures Laboratory Experiments for General Chemistry Harold R. Hunt, Toby F. Block, 1994 *English in Analytical Chemistry. Communicating about Methods & Techniques. Книга для студента* Надежда Зорина, Александра Соболева, 2022-10-19 **Scientific and Technical Aerospace Reports**, 1995-05 **General Chemistry** Hunt, 1976-06-07 **Laboratory Manual for Principles of General Chemistry** J. A. Beran, Mark Lassiter, 2022-08-16 Laboratory Manual for Principles of General Chemistry 11th Edition covers two semesters of a general chemistry laboratory program The material focuses on the lab experiences that reinforce the concepts that not all experimental conclusions are the same and depend on identifying an appropriate experimental procedure selecting the proper apparatus employing the proper techniques systematically analyzing and interpreting the data and minimizing inherent variables As a result of good data a scientific and analytical conclusion is made which may or may not be right but is certainly consistent with the data Experiments write textbooks textbooks don't write experiments A student's scientific literacy grows when experiences and observations associated with the scientific method are encountered Further experimentation provides additional cause effect observations leading to an even better understanding of the experiment The 11th edition's experiments are informative and challenging while offering a solid foundation for technique safety and experimental procedure The reporting and analysis of the data and the pre and post lab questions focus on the intuitiveness of the experiment The experiments may accompany any general chemistry textbook and are compiled at the beginning of each curricular unit An Additional Notes column is included in each experiment's Report Sheet to provide a space for recording observations and data during the experiment Continued emphasis on handling data is supported by the Data Analysis section **General, Organic, and Biological Chemistry Study Guide and Selected Solutions** Karen C. Timberlake, 2001-11 Keyed to the learning goals in the text this guide is designed to promote active learning through a variety of exercises with answers and mastery exams The guide also contains complete solutions to odd numbered problems Issues in Education by Subject, Profession, and Vocation: 2013 Edition, 2013-05-01 Issues in Education by Subject Profession and Vocation 2013 Edition is a ScholarlyEditions book that delivers timely authoritative and comprehensive information about Health Education Research The editors have built Issues in Education by Subject Profession and Vocation 2013 Edition on the vast information databases of ScholarlyNews You can expect the information about Health Education Research in this book to be deeper than what you can access anywhere else as well as consistently reliable authoritative informed and relevant The content of Issues in Education by Subject Profession

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Table of Contents Analysis Of An Unknown Salt Lab Report

1. Understanding the eBook Analysis Of An Unknown Salt Lab Report
 - The Rise of Digital Reading Analysis Of An Unknown Salt Lab Report
 - Advantages of eBooks Over Traditional Books
2. Identifying Analysis Of An Unknown Salt Lab Report
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Analysis Of An Unknown Salt Lab Report
 - User-Friendly Interface
4. Exploring eBook Recommendations from Analysis Of An Unknown Salt Lab Report
 - Personalized Recommendations
 - Analysis Of An Unknown Salt Lab Report User Reviews and Ratings
 - Analysis Of An Unknown Salt Lab Report and Bestseller Lists

5. Accessing Analysis Of An Unknown Salt Lab Report Free and Paid eBooks
 - Analysis Of An Unknown Salt Lab Report Public Domain eBooks
 - Analysis Of An Unknown Salt Lab Report eBook Subscription Services
 - Analysis Of An Unknown Salt Lab Report Budget-Friendly Options
6. Navigating Analysis Of An Unknown Salt Lab Report eBook Formats
 - ePub, PDF, MOBI, and More
 - Analysis Of An Unknown Salt Lab Report Compatibility with Devices
 - Analysis Of An Unknown Salt Lab Report Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Analysis Of An Unknown Salt Lab Report
 - Highlighting and Note-Taking Analysis Of An Unknown Salt Lab Report
 - Interactive Elements Analysis Of An Unknown Salt Lab Report
8. Staying Engaged with Analysis Of An Unknown Salt Lab Report
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Analysis Of An Unknown Salt Lab Report
9. Balancing eBooks and Physical Books Analysis Of An Unknown Salt Lab Report
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Analysis Of An Unknown Salt Lab Report
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Analysis Of An Unknown Salt Lab Report
 - Setting Reading Goals Analysis Of An Unknown Salt Lab Report
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Analysis Of An Unknown Salt Lab Report
 - Fact-Checking eBook Content of Analysis Of An Unknown Salt Lab Report
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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