

Stoichiometry

Solve the following problems:

1. Four moles of iron (II) sulfide added to seven moles of oxygen, produced two moles of iron (III) oxide and four moles of sulfur dioxide. (20pts)

a) Translate the problem to chemical equation.



Reactant side

4 Fe atoms

4 S atoms

7(2) =14 O atoms

Product side

4 Fe atoms

4 S atoms

2(3) +4(2) =14 O atoms

b) How many grams of oxygen will react with 2.0 grams of iron (II) sulfide?

Given:

m FeS = 2.0 g

Req'd. = m O₂

Solution:

molar mass of FeS = 88 g/mol

mole ratio = 4 mol FeS: 7 mol O₂

molar mass of O₂ = 32 g/mol

$$m_{O_2} = 2.0 \cancel{g_{FeS}} \left(\frac{1 \cancel{mol_{FeS}}}{88 \cancel{g_{FeS}}} \right) \left(\frac{7 \cancel{mol_{O_2}}}{4 \cancel{mol_{FeS}}} \right) \left(\frac{32 \cancel{g_{O_2}}}{1 \cancel{mol_{O_2}}} \right) = 1.3 g_{O_2}$$

2. Rusting involves the reaction of iron, Fe with oxygen, O₂ to form iron (III) oxide, Fe₂O₃. (20pts)

a) Write the balanced chemical equation.



Reactant side

4 Fe atoms

3(2) =6 O atoms

Product side

4 Fe atoms

2(3) =6 O atoms

b) What are the mole ratios of the product to each reactant and of the reactants to each other?

Reactant to each other:

4 mol Fe: 3 mol O₂

Product to each reactant:

2 mol Fe₂O₃:4 mol Fe

Answer Key To Stoichiometry Homework Problems

Jin-Ying Zhang



Answer Key To Stoichiometry Homework Problems:

Whispering the Techniques of Language: An Mental Journey through **Answer Key To Stoichiometry Homework Problems**

In a digitally-driven world where displays reign great and immediate interaction drowns out the subtleties of language, the profound strategies and mental nuances concealed within words frequently move unheard. However, situated within the pages of **Answer Key To Stoichiometry Homework Problems** a fascinating literary prize blinking with fresh emotions, lies an extraordinary quest waiting to be undertaken. Penned by a talented wordsmith, that wonderful opus encourages viewers on an introspective journey, softly unraveling the veiled truths and profound affect resonating within the fabric of each and every word. Within the emotional depths with this emotional evaluation, we will embark upon a honest exploration of the book is core themes, dissect its charming publishing style, and succumb to the powerful resonance it evokes serious within the recesses of readers hearts.

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