

# Pre-lab Questions for Experiment #5

## Rates of Chemical Reactions I: A Clock Reaction

1) Complete and balance the following reactions:



2) The rate law for the reaction



is  $\text{rate} = k[\text{H}_2][\text{NO}]^2$ . Which of the following mechanisms can be ruled out on the basis of the observed rate expression?

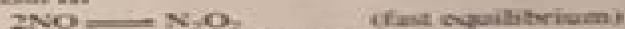
*Mechanism I*



*Mechanism II*



*Mechanism III*



3) Consider the reaction



From the following data obtained at a certain temperature, determine the order of the reaction and calculate the rate constant:

| [A] (M) | [B] (M) | Rate (M/s)            |
|---------|---------|-----------------------|
| 1.50    | 1.50    | $3.20 \times 10^{-1}$ |
| 1.50    | 2.50    | $3.20 \times 10^{-1}$ |
| 3.00    | 1.50    | $6.40 \times 10^{-1}$ |

4) The following gas-phase reaction was studied at  $290^\circ\text{C}$  by observing the change in pressure as a function of time in a constant-volume vessel:



Determine the order of the reaction and the rate constant based on the following data:

| Time (s) | Partial Pressure of $\text{ClCO}_2\text{CCl}_3$ (mmHg) |
|----------|--------------------------------------------------------|
| 0        | 15.76                                                  |
| 181      | 12.64                                                  |
| 513      | 8.73                                                   |
| 1164     | 4.44                                                   |

# Answer Of Pre Lab Rates Of Reaction

**Dr. K. N. Sharma, Dr. Subhash Chandra  
Rastogi, Er. Meera Goyal**

## Answer Of Pre Lab Rates Of Reaction:

**Integrated Approach to Coordination Chemistry** Rosemary A. Marusak, Kate Doan, Scott D. Cummings, 2007-03-07 Coordination chemistry is the study of compounds formed between metal ions and other neutral or negatively charged molecules This book offers a series of investigative inorganic laboratories approached through systematic coordination chemistry It not only highlights the key fundamental components of the coordination chemistry field it also exemplifies the historical development of concepts in the field In order to graduate as a chemistry major that fills the requirements of the American Chemical Society a student needs to take a laboratory course in inorganic chemistry Most professors who teach and inorganic chemistry laboratory prefer to emphasize coordination chemistry rather than attempting to cover all aspects of inorganic chemistry because it keeps the students focused on a cohesive part of inorganic chemistry which has applications in medicine the environment molecular biology organic synthesis and inorganic materials **Lab Manual Chemistry Class XII -by Dr. K. N. Sharma, Dr. Subhash Chandra Rastogi, Er. Meera Goyal (SBPD Publications)** Dr. K. N. Sharma, Dr. Subhash Chandra Rastogi, Er. Meera Goyal, 2021-07-03 Highly Useful for Various Engineering and Medical Competitive Examinations Chemistry Lemay, Prentice-Hall Staff, 1996 **Nuclear Science Abstracts**, 1971-05 Chemical Investigations Nancy Konigsberg Kerner, 1986 The Latest and Best of TESS, 1991 **Chemistry** McGraw-Hill Staff, 2001-07 **Experimental Organic Chemistry** Daniel R. Palleros, 2000-02-04 This cutting edge lab manual takes a multiscale approach presenting both micro semi micro and macroscale techniques The manual is easy to navigate with all relevant techniques found as they are needed Cutting edge subjects such as HPLC bioorganic chemistry multistep synthesis and more are presented in a clear and engaging fashion **Who's the New Kid in Chemistry?** John D. Butler, 2013-12-12 Who s the New Kid in Chemistry offers an unprecedented look at student engagement and teacher best practices through the eyes of an educational researcher enrolled as a public high school student Over the course of seventy nine consecutive days John D Butler participates in and observes Rhode Island 2013 Teacher of the Year Jessica M Waters s high school chemistry class documenting his experiences as they unfold Who s the New Kid in Chemistry is a compelling example of what can be accomplished when an educational researcher and teacher collaborate in the classroom This work includes a discussion on flexible homework assignments data driven instruction and thirty teacher best practices This book is an invaluable resource for teachers across all content areas masters and doctoral research method classes and future Teachers of the Year **Laboratory Manual on Biotechnology** P. M. Swamy, 2008

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In a global eaten by monitors and the ceaseless chatter of instantaneous interaction, the melodic splendor and psychological symphony produced by the written word often fade into the back ground, eclipsed by the relentless sound and interruptions that permeate our lives. But, located within the pages of **Answer Of Pre Lab Rates Of Reaction** a stunning fictional prize full of raw thoughts, lies an immersive symphony waiting to be embraced. Constructed by a wonderful musician of language, that charming masterpiece conducts viewers on an emotional journey, skillfully unraveling the hidden melodies and profound impact resonating within each cautiously constructed phrase. Within the depths of this touching review, we can discover the book is central harmonies, analyze their enthralling writing model, and surrender ourselves to the profound resonance that echoes in the depths of readers souls.

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