

◀ Homework Chapter 6

Exercise 6.39: Problems - The Mole Concept

Part A

How many carbon atoms are in a diamond (pure carbon) with a mass of 75 mg?

Express your answer using two significant figures.

$$N = 2.5 \cdot 10^{21} \quad \text{C atoms}$$

 You have already submitted this answer. Enter a new answer.
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Answers Mastering Chemistry Homework

Wolfgang Guggemos

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