# **Chemical Formula Writing Worksheet Solutions**

Write chemical formulas for the compounds in each box. The names are found by finding the intersection between the cations and anions. Example: The first box is the intersection between the "zinc" cation and the "chloride" anion, so you should write "ZnCl<sub>2</sub>", as shown.

Service	zinc	iron (II)	iron (III)	gallium	silver	lead (IV)
chloride	ZnCl <sub>2</sub>	FeCl <sub>2</sub>	FeCl <sub>3</sub>	GaCl <sub>3</sub>	AgCI	PbCl <sub>a</sub>
acetate	Zn(C <sub>2</sub> H <sub>1</sub> O <sub>2</sub> ) <sub>2</sub>	Fe(C <sub>2</sub> H <sub>2</sub> O <sub>2</sub> ) <sub>2</sub>	Fe(C <sub>2</sub> H <sub>3</sub> O <sub>2</sub> ) <sub>3</sub>	Ga(C <sub>2</sub> H <sub>3</sub> O <sub>2</sub> ) <sub>3</sub>	Ag C <sub>2</sub> H <sub>3</sub> O <sub>2</sub>	Pb(C <sub>2</sub> H <sub>1</sub> O <sub>2</sub> ) <sub>4</sub>
nitrate	Zn(NO <sub>3</sub> ) <sub>2</sub>	Fe(NO <sub>3</sub> ) <sub>2</sub>	Fe(NO <sub>3</sub> ) <sub>3</sub>	Ga(NO <sub>3</sub> ) <sub>3</sub>	AgNO <sub>3</sub>	Pb(NO <sub>3</sub> ) <sub>4</sub>
oxide	ZnO	FeO	Fe <sub>2</sub> O <sub>3</sub>	Ga <sub>2</sub> O <sub>3</sub>	Ag <sub>2</sub> O	PbO <sub>2</sub>
nitride	Zn <sub>3</sub> N <sub>2</sub>	Fe <sub>3</sub> N <sub>2</sub>	FeN	GaN	Ag <sub>2</sub> N	Pb <sub>3</sub> N <sub>4</sub>
sulfate	ZnSO <sub>4</sub>	FeSO <sub>4</sub>	Fe <sub>2</sub> (SO <sub>4</sub> ) <sub>3</sub>	Ga <sub>2</sub> (SO <sub>4</sub> ) <sub>3</sub>	Ag <sub>2</sub> SO <sub>4</sub>	Pb(SO <sub>4</sub> ) <sub>2</sub>

Write the formulas for the following compounds:

- copper (II) chloride CuCl<sub>2</sub>
- lithium acetate LiC<sub>2</sub>H<sub>3</sub>O<sub>2</sub>
- vanadium (III) selenide V<sub>2</sub>Se<sub>3</sub>
- 4) manganese (IV) nitride Mn<sub>3</sub>N<sub>4</sub>
- 5) beryllium oxide BeO
- sodium sulfate Na<sub>2</sub>SO<sub>4</sub>
- 7) aluminum arsenide AlAs
- potassium permanganate KMnO<sub>4</sub>
- 9) chromium (VI) cyanide Cr(CN)<sub>6</sub>
- 10) tin (II) sulfite SnSO<sub>3</sub>
- 11) vanadium (V) fluoride VF<sub>5</sub>
- ammonium nitrate NH<sub>4</sub>NO<sub>3</sub>

# **Answers For Ionic Formula Writing**

Peter J. Mikulecky, Chris Hren

#### **Answers For Ionic Formula Writing:**

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