Stoichiometry Lab: Vinegar and Baking Soda Do Not Write On This Sheet

<u>Purpose</u>: To predict the amount of Carbon Dioxide gas that should be produced in a chemical reaction; then calculate the % yield.

CH, COOH + NaHCO; ---> NaCH, COO + H,O + CO,

Materials: Baking Soda (NaHCO₃), Vinegar (CH₃COOH), and 2 plastic cups, scale.

<u>Procedure</u>: Find and record the mass of cup A. With cup A still on the scale, add approximately 10.0 g of baking soda to the cup. The mass does not have to be exact, as long as you carefully record your results.

Place cup B on the scale, weigh and record approximately 50.0 g of vinegar. The mass does not have to be exact, as long as you carefully record your results.

Slowly add vinegar to cup A until the reaction has stopped. DO NOT add all of the vinegar, just enough to complete the reaction. Reweigh and record both cup A and B. Calculate the mass of CO₂ that escaped.

Data:

- a. Mass of Cup A _____g
- b. Mass of Cup A and baking soda g
- c. Calculate mass of baking soda (b a) _____ g
- d. Mass of Cup B with vinegar g
- e. Mass of Cup B after reaction g
- f. Calculate mass of vinegar poured into Cup A (d e) g
- g. Mass of Cup A after reaction _____g
- b. Calculate mass of product after reaction (g a) ______g
- i. Calculate baking soda + vinegar (c + f) _____g
- j. Calculate mass of CO₂ lost (i h) _____g

Analysis Of Baking Soda Lab Answers

M Mark

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web Áni was one of 9 crew members onboard skalla grímur s viking ship they were all fleeing the norwegian king haraldur hárfagri harald fair hair like so many of the settlers of iceland the son of Áni was named Ögmundur and he lived at Ánabrekka Áni slopes when egill skallarímsson lived at borg

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