

# Stoichiometry

Solve the following problems:

1. Four moles of iron (II) sulfide added to seven moles of oxygen, produced two moles of iron (III) oxide and four moles of sulfur dioxide. (20pts)

a) Translate the problem to chemical equation.



**Reactant side**

4 Fe atoms

4 S atoms

7(2) = 14 O atoms

**Product side**

4 Fe atoms

4 S atoms

2(3) + 4(2) = 14 O atoms

b) How many grams of oxygen will react with 2.0 grams of iron (II) sulfide?

**Given:**

m FeS = 2.0 g

Req'd. = m O<sub>2</sub>

**Solution:**

molar mass of FeS = 88 g/mol

mole ratio = 4 mol FeS: 7 mol O<sub>2</sub>

molar mass of O<sub>2</sub> = 32 g/mol

$$m_{O_2} = 2.0 \cancel{g_{FeS}} \left( \frac{1 \cancel{mol_{FeS}}}{88 \cancel{g_{FeS}}} \right) \left( \frac{7 \cancel{mol_{O_2}}}{4 \cancel{mol_{FeS}}} \right) \left( \frac{32 \cancel{g_{O_2}}}{1 \cancel{mol_{O_2}}} \right) = 1.3 g_{O_2}$$

2. Rusting involves the reaction of iron, Fe with oxygen, O<sub>2</sub> to form iron (III) oxide, Fe<sub>2</sub>O<sub>3</sub>. (20pts)

a) Write the balanced chemical equation.



**Reactant side**

4 Fe atoms

3(2) = 6 O atoms

**Product side**

4 Fe atoms

2(3) = 6 O atoms

b) What are the mole ratios of the product to each reactant and of the reactants to each other?

**Reactant to each other:**

4 mol Fe: 3 mol O<sub>2</sub>

**Product to each reactant:**

2 mol Fe<sub>2</sub>O<sub>3</sub>: 4 mol Fe

# Answer Key To Stoichiometry Homework Problems

**Daniel C. Tofan**



## **Answer Key To Stoichiometry Homework Problems:**

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**Multiple Solution Methods**

**for Teaching Science in the Classroom** Stephen DeMeo, 2008 For the first time in science education the subject of multiple solution methods is explored in book form While a multiple method teaching approach is utilized extensively in math education there are very few journal articles and no texts written on this topic in science Teaching multiple methods to science students in order to solve quantitative word problems is important for two reasons First it challenges the practice by teachers that one specific method should be used when solving problems Secondly it calls into question the belief that multiple methods would confuse students and retard their learning Using a case study approach and informed by research conducted by the author this book claims that providing students with a choice of methods as well as requiring additional methods as a way to validate results can be beneficial to student learning A close reading of the literature reveals that time spent on elucidating concepts rather than on algorithmic methodologies is a critical issue when trying to have students solve problems with understanding It is argued that conceptual understanding can be enhanced through the use of multiple methods in an environment where students can compare evaluate and verbally discuss competing methodologies through the facilitation of the instructor This book focuses on two very useful methods proportional reasoning PR and dimensional

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**Chemistry Homework** Frank Schaffer Publications, Joan DiStasio, 1996-03 Includes the periodic table writing formulas balancing equations stoichiometry problems and more

*Basics for Chemistry* David A. Ucko, 2013-09-24 Basics of Chemistry provides the tools needed in the study of General Chemistry such as problem solving skills calculation methods and the language and basic concepts of chemistry The book is designed to meet the specific needs of underprepared students Concepts are presented only as they are needed and developed from the simple to the complex The text is divided into 18 chapters each covering some particular aspect of chemistry such as matter energy and measurement the properties of atoms description of chemical bonding study of chemical change and nuclear and organic chemistry Undergraduate students will find the book as a very valuable academic material

**General Chemistry Workbook** Daniel C. Tofan, 2010-07-28 This workbook is a comprehensive collection of solved exercises and problems typical to AP introductory and general chemistry courses as well as blank worksheets containing further practice problems and questions It contains a total of 197 learning objectives grouped in 28 lessons and covering the vast majority of the types of problems that a student will encounter in a typical one year chemistry course It also contains a fully solved 50 question practice test which gives students a good idea of what they might expect on an actual final exam covering the entire material

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